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## EVIDENCING THE CHEMICAL DEGRADATION OF A HYDROPHILIZED PES ULTRAFILTRATION MEMBRANE DESPITE PROTEIN FOULING

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### KEYWORDS

PES, PVP, ultrafiltration, proteins, ageing, hypochlorite, FTIR-ATR

### ABSTRACT

Hydrophilisation of polyethersulfone (PES) based membrane is often achieved by addition of polyvinylpyrrolidone (PVP) leading to a physical blend of the two polymers. This paper shows that the most commonly used membrane for UF in dairy industry is a PES/PVP based one. Nevertheless if hydrophilisation limits the organic fouling, PVP is also the Achilles heel of these membranes. It is particularly true when membranes are exposed to hypochlorite as it is the case for cleaning/disinfection steps. Evidencing the disappearance of PVP from a pristine PES/PVP membrane can be easily achieved by FTIR-ATR analyses. But when one wants to study the ageing of a membrane used in UF it gets more complicated: regardless of the cleaning efficiency the membrane always remains fouled by some proteins. As both PVP and proteins own chemical bounds leading to absorption at the same wavenumber in FTIR-ATR, it thereby prevents the easy highlighting of the PVP degradation. The aim of this paper is to propose a simple treatment of raw FTIR-ATR spectra to dissociate these two contributions, allowing consequently the study of the degradation of a fouled membrane. Then the procedure is applied to a real case study on a spiral membrane.

## 1. INTRODUCTION

Fouling systematically occurs during ultrafiltration (UF) as it is observed when filtering skim milk by spiral membranes based on polyethersulfone (PES, **Figure 1**) [1-4]. The membrane composition is of course an important parameter as the physico-chemical interactions between the skim milk components and the membrane material play a crucial role on the membrane global fouling [5]. In fact, PES membranes also contained a minor amount of additives that are added to increase the membrane hydrophilic character, but generally these additives are not clearly mentioned by the membrane providers. Polyvinylpyrrolidone (PVP) (**Figure 1**) is one of the most famous used additive and it is quite logical to suspect its presence in a hydrophilised PES membrane. But it must generally be proved as it will be done in this study as preliminary investigation (see results).

Nowadays, it is quite well understood that hydrophilisation limits the membrane fouling by organic compounds such as proteins. Nevertheless, it is also known that PVP is the Achilles heel of these membranes. This is particularly true when they are exposed to sodium hypochlorite (NaOCl) as it is the case for daily cleaning/disinfection steps at industrial scale.

To go ahead in the understanding in PES/PVP membrane stability, several teams working in membrane food or water applications have chosen to study variations of flux and selectivity after immersion of pristine membranes in various NaOCl solutions. For such purposes they used more or less complex degradation protocols [2, 6, 8-22].

Dealing with the membrane material degradation, in our opinion one of the most comprehensive fundamental results is obtained by infra-red analyses (FTIR). Infra-red spectroscopy is a technique only able to evidence polar bounds existing in polymers. Nevertheless it must be underlined that these studies are achieved on pristine membranes that were voluntarily aged in sodium hypochlorite solutions, the pH of which is generally set in the range 8.0 – 9.0. NaOCl disinfecting solutions having such pHs are known to lead to severe degradation of PES/PVP membranes [see for instance 18, 19 and references cited herein].

In the case of PES/PVP membranes a special attention must be paid to an absorption band of the FTIR-ATR spectrum located close to a wavenumber at  $1661\text{ cm}^{-1}$  and attributed to the C=O amide bound of PVP. To put it in a nutshell, the PVP C=O bound progressively disappears during the membrane ageing provoked by NaOCl treatments. **Figure 2** shows one

of the mechanism suggested in literature to explain the C=O disappearance that can be understood as the opening of the PVP ring. This is significantly different from a total removal of the PVP or of its degradation compounds.

Moreover, another band located at  $1030\text{ cm}^{-1}$  progressively appears on the FTIR-ATR spectrum and is commonly attributed to an evolution of the PES skeleton itself. Nevertheless there is still controversy about the exact meaning of this band, sometimes assigned to a sulfonic acid group ( $\text{SO}_3\text{H}$ , **Figure 3a**, [14]) and recently to a phenol formation (OH on phenyl ring, **Figure 3b**, [18]). It is out of the scope of this paper to discuss the validity of these two proposals and in the following we will only discuss degradation of PES skeleton evidenced by the  $1030\text{ cm}^{-1}$  band without any other comments.

To show the evolution of the PES skeleton is quite easy on an aged membrane that have been used at industrial scale because no band overlapping have been evidenced with the fouling (**Figure 4**) and the  $1030\text{ cm}^{-1}$  is a pure signal revealing the PES degradation. For a more detailed presentation of the membrane used at industrial scale and called here “Membrane U” see [1].

The disappearance of PVP from a pristine PES/PVP membrane can be easily achieved by FTIR-ATR analysis, but only a single band can be used for this purpose: those located at  $1661\text{ cm}^{-1}$ . But when one wants to study the ageing of a membrane used in UF it gets more complicated because of the presence of fouling as it will be explained below.

Dealing with PES/PVP membranes fouled during UF of skim milk, it must be underlined that only a part of the whole fouling is irreversible (not removed by a simple water rinsing) and this part is known to be made of proteins only, in the particular case of organic membranes [3-4]. This irreversible protein deposit represents the target of the cleaning/disinfecting operation that immediately follows the production step. Classically, at industrial scale, organic UF membranes fouled by skim milk are cleaned according to a procedure involving a cascade of 3 solutions set at  $50^\circ\text{C}$ : alkaline cleaning at pH 11.5/12.0 with a formulated detergent containing surfactants, then acid cleaning generally with nitric acid at pH 1.6 and finally an alkaline and chlorine disinfecting step using sodium hypochlorite (bleach) set at 150-200 ppm in total free chlorine (TFC) [1, 3, 4]. In our previous studies we have explained how such a cascade can be simplified by the cancellation of the acid step [7]. Nevertheless for food security purposes it is not possible to withdraw the disinfecting step. Physico-chemical



analyses of such industrial membranes autopsied at the end of their service life highlight presence of a residual protein fouling even after a final cleaning procedure [1].

Proteins own a specific bound called “amide I” (C=O vibration of the amide group, -CO-NH) that is superimposed to the PVP C=O bound (the only one that can be used to deal with the PVP degradation). Consequently, an analytical difficulty must be overcome when fouling is made of proteins (**Figure 5**) to unambiguously evidence modifications of the membrane due to fouling occurrence or to membrane material degradation or a combination of the two.

So in this paper we propose an approach to treat a FTIR-ATR spectrum in order to dissociate the contributions of PVP and proteins to the global band obtained on the raw spectrum of a fouled membrane. For the demonstration pristine membranes at 3 degradation states are prepared and fouled with proteins. Then the methodology is applied to the analysis of a spiral membrane (6.7 m<sup>2</sup>) used at laboratory scale alternatively in UF of skim milk and UF of sodium hypochlorite aiming at evidencing a possible heterogeneity of the membrane degradation with respect to the distribution of fouling in this complex geometry.

## 2. THEORETICAL PART

On the FTIR-ATR spectra of a PES/PVP pristine membrane (here HFK-131 membrane) all bands can be assigned to PES except that located at 1661 cm<sup>-1</sup> which can be attributed to the C=O bound of PVP. **Figure 4** shows the attributions in the 1770-700 cm<sup>-1</sup> range that is the range of interest for the following discussion. For sake of comparison, the spectrum of a membrane already used at industrial scale (Membrane U) is also given. Membrane U is at the end of its service life that corresponds to 8,000 h of skim milk UF. A final and very efficient cleaning has been performed before the FTIR-ATR spectrum registration. As explained in the introduction, clearly the PVP C=O band has disappeared and the new 1030 cm<sup>-1</sup> band has appeared on the spectrum.

However, when the membrane is fouled by milk proteins and when the cleaning is not sufficiently efficient to remove the proteins from the membrane, the PVP C=O band and the amide I band of proteins are overlapped. This analytical difficulty would be more and more frequent with the membrane ageing because the fouling tends to increase with the membrane age and the cleanability seems to decrease accordingly. **Figure 5** shows the raw spectra obtained for a pristine membrane and for a protein fouled one. If the protein deposit is very

high (see results) then an aged and fouled membrane can exhibit a C=O band of greater intensity than that of the pristine membrane itself. Consequently it becomes impossible to deal with a possible degradation of PVP from the membrane without any treatment of the FTIR-ATR spectrum.

We propose to establish a mathematical treatment of the spectrum in order to dissociate the contribution of these two C=O bands. Moreover, this paper aims at developing a simple approach that can be applied without any sophisticated and costly software able to perform band deconvolution.

Proteins also lead to a second characteristic absorption band called amide II band that corresponds to the superimposition of vibrations due to CN+NH bounds of the amide group.

Knowing that the protein amide II band, located close to  $1539\text{ cm}^{-1}$ , have no overlapping with the pristine membrane bands, neither PES nor PVP, we have already developped a method of quantification of proteins directly on the HFK-131 membrane. This analysis is based on the  $H_{1539}^{\text{protein amide II}} / H_{1240}^{\text{PES}}$  ratio (with  $H_i^X$ : band height corresponding to the absorbance intensity at a given  $i$  wavenumber ( $w_i$ ) and corresponding to X material or functional group) [4].

The first step of the approach proposed in this paper is based on the simple fact that on a fouled membrane, absorption measured at  $1661\text{ cm}^{-1}$  ( $H_{1661}^{\text{raw spectrum}}$ ) corresponds to the superimposition of absorbance due to PVP ( $H_{1661}^{\text{PVP}}$ ) and absorbance due to protein amide I vibration ( $H_{1661}^{\text{protein amide I}}$ ):

$$H_{1661}^{\text{raw spectrum}} = H_{1661}^{\text{PVP}} + H_{1661}^{\text{protein amide I}} \quad (\text{eq. 1})$$

The new idea of this paper consists in the determination of the protein amide I intensity at  $1661\text{ cm}^{-1}$  ( $H_{1661}^{\text{protein amide I}}$ ) from that of the amide II band ( $H_{1539}^{\text{protein amide II}}$ ).

As mentioned previously, these two bonds belong to the same amide group, namely N-C=O, but correspond to different vibration modes. Consequently it is not creasy to imagine that the intensity of the two FTIR bands of the amide group of proteins are correlated. Assuming this idea, we have tried to evidence the possible relationship between the two heights of these bounds without any other assumption. The experimental developments detailed in the results

show that the relationship between the two heights exists. Moreover experimental results evidence that, regardless of the chosen protocol, the relationship is linear (see below). For sake of simplification the general form of this relationship is consequently given as that of a straight line (**equation 2**).

$$H_{1661}^{\text{protein amide I}} = \alpha H_{1539}^{\text{protein amide II}} + \beta \quad (\text{eq. 2})$$

Where  $\alpha$  and  $\beta$  are coefficients to be determined.

A main bottleneck of the approach is the experimental determination of  $\alpha$  and  $\beta$ . In the results we will discuss two experimental approaches to determine these coefficients. One is based on spectra acquired for proteins in aqueous solutions. The other involved treatment of spectra acquired with more or less aged membranes on which controlled protein deposit are made. It is explained in the results why only one of these two approaches can be validated.

The second key point is the experimental validation of **equation 3** that can be easily drawn from the previous ones.

$$H_{1661}^{\text{raw spectrum}} / H_{1240}^{\text{PES}} = (H_{1661}^{\text{PVP}} / H_{1240}^{\text{PES}}) + (H_{1661}^{\text{protein amide I}} / H_{1240}^{\text{PES}}) \quad (\text{eq. 3})$$

It requires having PES/PVP membranes with various amount of PVP. In this paper we have used pristine and aged membranes specially prepared to cover a wide range of membrane degradation up to a PVP height divided by two with respect to its initial height.

**Figure 6** summarizes the global approach.

### 3. EXPERIMENTAL

#### 3.1.1. Water, skim milk and solutions

Water used either for solution preparation and membrane rinsing is deionised and 1  $\mu\text{m}$  filtered. Its conductivity is always lower than 1  $\mu\text{S.cm}^{-1}$ .

The skim milk used is a commercial one (UHT, Lait de Montagne, Carrefour, France) containing an average of 32 g.L<sup>-1</sup> proteins and 48 g.L<sup>-1</sup> carbohydrates (mainly lactose) and only traces of lipids (< 0.5 %).

Protein solutions are prepared from a powder containing whey proteins (PS90 Protarmor, Saint Brice en Coglès, France) dissolved in deionised water. A set of 6 concentrations is used: 0, 5, 10, 20, 30 and 50 g.L<sup>-1</sup>.

NaOCl solutions at 400 ppm in total free chlorine (TFC) are prepared from commercial bleach solution provided by Lacroix (eau de Javel, bleach at 96,000 ppm TFC, France). The pH is adjusted to  $8.0 \pm 0.1$  by addition of HCl of analytical grade (Acros).

A formulated alkaline solution, namely P3-ultrasil 10 provided by Ecolab (Issy les Moulineaux, France) is used for membrane cleaning. It is set at 4 g.L<sup>-1</sup> that corresponds to a natural pH of 12.0.

### 3.1.2. Membranes

The membrane used is named HFK-131. It is a UF membrane provided by Koch (USA) and commonly used at industrial scale for UF of skim milk and of both acid and rennet whey. This membrane represents about 70 % of the world market for these specific applications. It is a PES based membrane of 5-10 kg mol<sup>-1</sup> molecular weight cut off according to the provider.

We have studied this membrane to prove that it is a PES/PVP one. Among experimental data allowing to draw this conclusion are:

- (i) the presence of the 1661 cm<sup>-1</sup> band on the FTIR-ATR spectra (**Figure 4, Figure 5**). Moreover, we had also checked that the HFK-131 spectrum is similar to that of a dense film we had prepared, that was made from PES and PVP physical blend (95/5) [6].
- (ii) the presence of nitrogen evidenced by nano-SIMS measurements (**Figure 7**). SIMS is only able to evidence nitrogen atoms bounded to carbon further called organic nitrogen (CN group).

### 3.1.2.1. Spiral membrane

The spiral membrane is chosen in order to have a filtering area of several square meters (6.7 m<sup>2</sup>, 4333 K131 VYV module). The membrane (further called CIP-3, KMS K131V 8879759136) is made of 4 double sheets of membranes wound around a permeate collector tube of 33 inches length as schematically explained in **Figure 8**.

### 3.1.2.2. Flat membrane coupons: fouling of pristine and aged membranes

Several flat membranes of 127 cm<sup>2</sup> (or sometimes less) are sampled in a second pristine spiral membrane (further called cut-CIP-1). These membrane coupons are used for FTIR-ATR analyses performed to validate the methodological approach explained in the theoretical part. Prior to be used these samples are carefully rinsed in water to remove the preservative (the rinsing efficiency is systematically checked by FTIR-ATR).

#### - Ageing of membrane coupons

The ageing protocol is inspired from the one established in our laboratory and already described in [6]. Six membrane coupons (6 cm x 3 cm) are soaked in a NaOCl solution set at 400 ppm TFC and pH 8.0 (V= 26.3 mL, surface/volume = 14.6 L.m<sup>-2</sup>) under micro-waves (pulsed microwaves at 255 W) during 210 or 480 min. The micro-waves help to accelerate the chemical degradation when compared to simple soaking in the same solution. Thus depending on the treatment duration different ageing states are reached; it is shown from the PVP C=O height decrease. In order to avoid impact of NaOCl concentration during the ageing process, the NaOCl solution has been regularly replaced by a similar volume of a fresh solution. Some characteristics of aged membranes are given **Table 1**.

#### - Protein fouling of membrane coupons

The protocol is the same when using pristine and aged membranes. A membrane coupon (3 cm x 6 cm) is immersed under gentle stirring at room temperature in 100 mL of a protein solution during 2 h. Proteins are set at different concentrations (0, 5, 10, 20, 30 and 50 g.L<sup>-1</sup>). After dripping, membranes are carefully dried in a desiccator under dynamic vacuum. FTIR-

ATR spectrum is registered on each dried coupon before and after the protein deposition. Protein quantification is achieved using **equation 4** (see below). The amount of proteins on obtained deposit are shown **Table 1** for the pristine membrane and the two aged membranes.

### 3.1.3. UF with the spiral membrane

#### 3.1.3.1. Skim milk UF

A pristine spiral membrane is installed on a pilot provided by TIA (TIA 3093, Bollène, France) and already described in [24]. The fouling of the spiral membrane is achieved by UF of 24 L of skim milk at 50°C in batch mode at volume reduction ratio VRR= 1, meaning here that both the retentate and the permeate are fully recycled in the feed tank. During UF the feed flow rate is constant. It is set at  $9.5 \text{ m}^3 \cdot \text{h}^{-1}$  that corresponds to a cross-flow velocity of about  $0.3 \text{ m} \cdot \text{s}^{-1}$ .

The transmembrane pressure (TMP) is regularly increased from 1.5 bar at start to 4.0 bar at the end of the UF run. This last TMP value allows to reach the limiting flux. Each TMP increase is achieved when a plateau value of flux is reached at the previous TMP. The UF duration is close to 5.5 h for one UF run. The membrane flux (J) is followed all over the skim milk UF.

The average TMP is classically calculated as the mean value when considering the membrane inlet and outlet TMPs. Nevertheless, it must be kept in mind that the pressure drop along the membrane element is 2.2 bar regardless of the average TMP. This is not a negligible value by comparison to the average TMP itself. In the following an estimation of the local TMP is made considering a linear pressure drop along the membrane element, according to a calculation proposed in [23].

At the end of skim milk UF, the membrane is extensively rinsed by water. Then it is cleaned at 2 bar, 50°C during 1 h by 25 L of P3-ultrasil 10 formulated solution. This protocol allows the water flux recovery of the pristine membrane.

#### 3.1.3.2. Sodium hypochlorite UF for membrane ageing

In order to accelerate the spiral membrane ageing, UF of sodium hypochlorite solution at 400 ppm TFC and pH 8.0 is performed at 2.0 bar and 50°C. The treatment duration can vary and is summarized as a cumulative dose of hypochlorite expressed according to **equation 5**.

$$\text{NaOCl dose membrane exposition (ppm.d)} = 400 \text{ ppm} \times \text{number of days} \quad (\text{eq. 4})$$

From a practical point of view, 3 UF of skim milk are consecutively performed on the pristine membrane. After alkaline cleaning the membrane is then aged in NaOCl up to a cumulative dose of 400 ppm.d. Then another UF of skim milk is performed. After a second alkaline cleaning the membrane is aged in NaOCl up to a cumulative dose of 2,000 ppm.d. Then another UF of skim milk is performed. After cleaning the membrane is aged up to a cumulative dose of 2,800 ppm.d and a final skim milk UF is performed. Finally, the membrane is carefully rinsed by water but not cleaned. Then the membrane is autopsied.

According to the provider this membrane is guaranteed up to a 5,000 ppm.d NaOCl dose but nothing is clearly explained about the way to reach this dose. Nevertheless, we can suggest that the autopsied CIP-3 membrane could be equivalent to a membrane close to half time of a normal service life at industrial scale.

#### 3.1.4. FTIR-ATR on membranes for protein quantification purpose

The membrane samples are carefully dried under dynamic vacuum before registration to remove water (systematically checked in the 3500 cm<sup>-1</sup> region).

For sake of generalization, three different FTIR-ATR spectrometers have been used in this study. Their characteristics are shown in **Table 2** as well as the standard conditions used for spectra acquisition. It has been checked that the spectra acquired on each spectrometer are similar when dealing with the wavenumbers but the raw intensity of each band is different. This is probably due to differences in the ATR-accessory crystal (chemical nature and reflection number). Nevertheless it has also been checked that the relative ratio of two band heights (for instance:  $H_i/H_{1240}^{\text{PES}}$ ) are the same regardless of the spectrometer used (see **Figure A1** in Appendix 1).

### 3.1.4.1. Quantification of residual proteins directly on membranes

The initial calibration has been established with spectrometer 1 [3, 4, 6] but as explained above can be used with spectrometer 2 (see **Figure A1** in Appendix 1) and spectrometer 3. The quantification is possible in a wide range of fouling from 1 to 350 µg of proteins per square centimetre of membrane (geometric area) with a precision of 1 µg.cm<sup>-2</sup> (**equation 4**). 19 samples of fouled pristine membranes have been used to establish the equation, allowing to reach  $r^2 = 0.997$ .

$$H_{1539}^{\text{protein amide II}} / H_{1240}^{\text{PES}} - H_{2060-2240}^{\text{baseline}} = 0.0034 [P] \quad (\text{eq. 4})$$

With:

[P]: the protein concentration in µg.cm<sup>-2</sup>

H<sub>i</sub><sup>X</sup>: band height corresponding to the absorbance intensity at a given wavenumber (w<sub>i</sub>) and corresponding to X material or functional group. (see **Appendix 1**).

H<sub>2060-2240</sub><sup>baseline</sup>: the average height of the baseline measured in the 2060-2240 cm<sup>-1</sup> range of wavenumbers. Value equal to 0.0165.

#### - Specific procedure for the spiral membrane

After the final fouling by skim milk up to 4.0 bar and rinsing by deionised water, the spiral membrane (CIP-3, 6.7 m<sup>2</sup>) has been cut in 324 pieces of about 127 cm<sup>2</sup> area each.

Each flat sample is then analyzed with spectrometer 1. Only the spectrum of the sample center is registered. Nevertheless, each 127 cm<sup>2</sup> sample could be divided in 9 smaller pieces of about 10 cm<sup>2</sup> (corresponding to the ZnSe crystal area), but it has been checked on a whole membrane sheet of about 1 m<sup>2</sup> that the variations are not significant and that a single result seems quite acceptable for the followed purpose.

This autopsy allows establishing firstly the mapping of the irreversible fouling and secondly that of the degradation state of the membrane by the mean of measurements at 1661 cm<sup>-1</sup> (PVP) and 1030 cm<sup>-1</sup> (PES skeleton).

#### - Specific procedure for the flat membrane coupons



The small flat membrane coupons have been analyzed with spectrometer 2. It was used to establish the protocol exposed in the theoretical part above.

### 3.1.5. FTIR-ATR on membranes for evidencing of membrane degradation

Firstly the spectrum of each membrane coupon, in pristine form, is registered. Thus  $H_{1240}^{PES}$  and  $H_{1661}^{PVP}$  is measured allowing the calculation of the  $H_{1661}^{PVP}/H_{1240}^{PES}$  ratio for each sample.

Secondly, some membrane coupons are submitted to a more or less intensive ageing procedure according to the protocols described in paragraph 3.1.2. Then the spectrum of each aged membrane coupon is registered. Thus  $H_{1240}^{PES}$  and  $H_{1661}^{PVP}$  is measured allowing the calculation of the  $H_{1661}^{PVP}/H_{1240}^{PES}$  ratio for each sample before fouling. Of course it has been checked that the PVP band is modified when compared to that measured before the ageing treatment. These values obtained after ageing are then used as reference values to be compared to values obtained after fouling on aged membranes.

Finally, membranes are fouled with protein solutions according to the protocol described in paragraph 3.1.2. FTIR-ATR spectra are then registered and  $H_{1661}^{raw\ spectra}$ ,  $H_{1240}^{PES}$  and  $H_{1661}^{protein\ amide\ II}$  are determined for the registered spectrum of each membrane coupon after fouling allowing the acquisition of the raw spectra of aged then fouled membranes.

The following ratios,  $H_{1661}^{raw\ spectra}/H_{1240}^{PES}$  and  $H_{1661}^{protein\ amide\ II}/H_{1240}^{PES}$  can easily be deduced. Then  $H_{1661}^{protein\ amide\ I}/H_{1240}^{PES}$  of the fouled membrane is drawn from **equation 3** with the  $H_{1661}^{raw\ spectra}/H_{1240}^{PES}$  ratio of the fouled membrane and the  $H_{1661}^{PVP}/H_{1240}^{PES}$  ratio of the same coupon before fouling.

Finally the  $H_{1661}^{protein\ amide\ I}/H_{1661}^{protein\ amide\ II}$  ratio is obtained from the  $(H_{1661}^{protein\ amide\ I}/H_{1240}^{PES})$  to  $(H_{1661}^{protein\ amide\ II}/H_{1240}^{PES})$  ratio.

### 3.1.6. FTIR-ATR on protein solutions

Spectra of 1  $\mu$ L of the several protein solutions have been registered with spectrometer 2.

The obtained spectrum of a solution (Protein raw spectrum) is the superimposition of the water spectrum and of the protein one. But because the main component of a protein solution is water that have a band located at  $1637\text{ cm}^{-1}$  which overlaps the amide I band of proteins, a treatment of the raw spectrum is needed to put in evidence amide I and amide II bands of proteins (**Equation 6**, see **Appendix 2** for details). After treatment the spectrum is called Protein difference spectrum

$$\text{Protein difference spectrum} = \text{Protein raw spectrum} - \gamma \text{water spectrum} \quad (\text{eq. 6})$$

With:

$\gamma$  coefficient to be determined for each protein spectrum. The determination of  $\gamma$  is based on the cancellation of the OH band of water located close to  $3300\text{ cm}^{-1}$ . There is no significant overlapping of the  $3300\text{ cm}^{-1}$  water band with protein bands in this region, because unsaturated C-H bonds of proteins are located below  $3000\text{ cm}^{-1}$ . The determination of  $\gamma$  is the result of tests and errors with different values close to 1.000 with the objective to obtain a region close to  $3300\text{ cm}^{-1}$  as flat as possible in the Protein difference spectrum that is the considered criterion to prove the quality of the spectrum treatment. Moreover, for similar reason, the  $2300\text{-}1800\text{ cm}^{-1}$  region must also be as flat as possible on the difference spectrum (see **Appendix 2** for details). From a practical point of view, in this study  $\gamma$  slightly varies around  $0.8 \pm 0.2$ .

## 4. RESULTS AND DISCUSSION

### 4.1. FTIR-ATR study of estimation of protein amide I band from protein amide II band

As already explained the first bottleneck of the procedure (**Figure 6**) consists in the estimation of protein amide I intensity from that of protein amide II on the FTIR-ATR spectrum. Two approaches have been tested that are detailed in the following.

#### 4.1.1. Proteins in solution

Spectra of  $1\text{ }\mu\text{L}$  of several protein solutions have been registered with spectrometer 2. Spectra have then been post-treated according to the procedure described in paragraph 3.1.6.

The results evidence that the  $H_{1661}^{\text{protein amide I}} / H_{1539}^{\text{protein amide II}}$  ratio decreases linearly with protein concentration according to **equation 7**.

$$H_{1661}^{\text{protein amide I}} / H_{1539}^{\text{protein amide II}} = - 0.06 \times [P'] + 3.6 \quad \text{with } r^2 = 0.95 \quad (\text{eq. 7})$$

With:

[P'] the concentration of proteins in g.L<sup>-1</sup>

This linear trend suggests a structural evolution of proteins with dilution.

#### **4.1.2. Proteins adsorbed on flat membrane**

Spectra of voluntary aged and fouled membrane coupons have been registered with spectrometer 1. Then spectra have been post-treated according to the procedure described in paragraph 3.1.5.

**Table 3**, **Table 4** and **Table 5** present FTIR-ATR data before and after fouling by proteins on the pristine membrane and the two aged ones.

Whatever the fouling amount and the membrane ageing state, the  $(H_{1661}^{\text{protein amide I}} / H_{1240}^{\text{PES}})$  ratio is always proportional to the  $(H_{1539}^{\text{protein amide II}} / H_{1240}^{\text{PES}})$  ratio; this means that  $\beta=0$  in **equation 2**.

Moreover, the  $(H_{1661}^{\text{protein amide I}} / H_{1539}^{\text{protein amide II}})$  ratio, and thus  $\alpha$  term of **equation 2**, is the same for the three aged state membranes in a wide range of protein deposit amount. (**Table 6**).

#### **4.1.3. Conclusion on the two approaches**

Estimation of the height of protein amide I band from the height of protein amide II band can be achieved from spectra acquired with proteins in solution and with proteins adsorbed on membranes. Nevertheless the conclusions are clearly different: in solution the ratio of the two band heights varies whereas it remains constant for proteins adsorbed on membranes.

In solution, this result suggests a structural evolution of proteins with dilution likely due to a competition between hydrogen bond due to protein-protein interactions and those due to protein-water interactions.

On membranes, adsorption is mainly due to hydrophobic interactions that lead to similar structures of adsorbed proteins, regardless of the amount that always remains low.

Finally the proposed approach can be validated by using calibration obtained with protein adsorbed on membranes because it is a closer situation to that of adsorbed proteins during UF (**Table 6**).

## 4.2. Spiral membrane degradation state

The degradation state is discussed from both flux measurements and FTIR-ATR spectra.

### 4.2.1. UF of skim milk

For pristine CIP-3 membrane, the permeability to water at 50°C is  $59.8 \text{ L.h}^{-1}.\text{m}^{-2}.\text{bar}^{-1}$ .

Three UF of skim milk have been performed on the pristine membrane with an alkaline cleaning step between each of them. **Figure 9** shows the excellent reproducibility.

After a sufficient dose of NaOCl received by the membrane, fluxes progressively increase, highlighting, on a hydrodynamic point of view, that membrane degradation occurs. It is also evidenced that more than the limiting flux the critical flux also varies with the membrane ageing (see [24] and references cited herein for a discussion dealing with limiting and critical fluxes with the pristine membrane in skim milk).

### 4.2.2. Mapping of the protein irreversible deposit

After the last skim milk UF (NaOCl dose set at 2,800 ppm.d) the membrane is rinsed and autopsied. **Figure 10** shows the distribution of the residual protein amount according to the location on the eight membrane sheets (supporting values are given **Figure A4-1** in **Appendix**

4). The zone closed to the permeate collector axis is often more fouled. An average value is calculated on all the membrane and corresponds to  $50 \pm 9 \mu\text{g.cm}^{-2}$ .

#### 4.2.3. PVP distribution on aged and fouled membrane

**Figure 11** shows the mapping of the  $(H_{1661}^{\text{raw spectra}}/H_{1240}^{\text{PES}})$  ratio as determined from raw spectra obtained by FTIR-ATR. An average ratio is calculated on the whole membrane corresponding to  $0.24 \pm 0.03$  (RSD = 11 %, the supporting values can be seen in **Figure A4-2** in **Appendix 4**).

**Figure 12** shows the mapping of the  $(H_{1661}^{\text{PVP}}/H_{1240}^{\text{PES}})$  ratio as determined from raw spectra obtained by FTIR-ATR and applying **equation 3**. An average ratio is calculated on the whole membrane corresponding to  $0.08 \pm 0.02$  (RSD = 18 %, the supporting values can be seen in **Figure A4-3** in **Appendix 4**). The PVP amount can be roughly considered as constant, highlighting a regular, but not total, disappearance of PVP on the whole membrane. This value is between that of the pristine membrane (0.14) and that of the industrial aged membrane (membrane U after extensive cleaning, 0.06).

In this spiral membrane, the protein amount varies from 10 to  $80 \mu\text{g}$  of proteins per  $\text{cm}^2$  but most values are in the range  $40\text{-}60 \mu\text{g.cm}^{-2}$ : fouling is not homogeneous whereas the PVP amount appears roughly constant. Consequently it is not possible to evidence if the protein fouling plays a protective role toward the PVP degradation. Nevertheless it is not possible to conclude on the absence of this protection.

#### 4.2.4. PES degradation on aged and fouled membrane

PES degradation has been checked from the  $1030 \text{ cm}^{-1}$  band increase. **Figure 13** shows the mapping of the  $(H_{1030}/H_{1240}^{\text{PES}})$  ratio as determined from raw spectra obtained by FTIR-ATR. An average ratio has been calculated on the whole membrane corresponding to  $0.13 \pm 0.06$  even if the relative standard deviation is very high (RSD = 45 %, the supporting values can be seen in **Figure A4-4** in **Appendix 4**). Consequently, this degradation state cannot be considered as constant and further investigations are needed to understand the origin of this observation. Nevertheless all these values are significantly lower than that of the industrial aged membrane (Membrane U after extensive cleaning, 0.29).

For similar reasons, it is not possible to evidence if the protein fouling plays or not a protective role toward the PES degradation.

Prulho et al. [19] and Pellegrin et al. [20] suggested a link between PVP degradation and PES degradation. These authors believe that some degradation compounds of PVP obtained from reaction involving NaOCl can be active organic radicals. These radicals would be able to attack the PES skeleton and lead to the 1030 cm<sup>-1</sup> band. So we tried to find correlation between PVP disappearance ( $H_{1661}^{PVP}/H_{1240}^{PES}$ ) and PES degradation ( $H_{1030}/H_{1240}^{PES}$ ). But no particular relationship was found (see **Figure A3-1** in Appendix 3). Our results were not able to confirm nor infirm the assumption of these authors.

## CONCLUSION

This paper proposes and validates an approach to evaluate the degradation state of a PES/PVP membrane even when the membrane is fouled by proteins.

The procedure is based on FTIR-ATR technique and can be used with spectrometers of different characteristics aiming at a generalization of these results.

The approach is then successfully applied to a case study of a spiral membrane voluntarily aged on a UF pilot by filtration of a 400 ppm NaOCl pH 8.0 solution alternatively with UF of skim milk.

Mapping of the irreversible fouling distribution in this membrane aged at 2,800 ppm.d NaOCl dose has been performed. The results demonstrate the existence of a protein distribution that is different from that observed at lower TMP with a pristine membrane [24]. The PVP remaining amount is quite regular showing a partial degradation but not a total disappearance. The PES skeleton is also degraded but in a more irregular way. An attempt of correlation between protein fouling and PVP disappearance or PES degradation failed. Nevertheless it is not possible to draw conclusion about the efficiency or the non-efficiency of protection of the fouling layer toward the whole process of membrane degradation. Moreover, it has not been possible to evidence a link between PVP disappearance and PES skeleton degradation.

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ACCEPTED MANUSCRIPT

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**Table captions**

**Table 1.** Conditions of membrane accelerated ageing, FTIR-ATR characteristics and protein amount further adsorbed on (aged) membrane surface.

\* $H_{1661}^{PVP}$  is the height of the band located at  $1661\text{ cm}^{-1}$  and attributed to PVP in the pristine membrane whereas  $H_{1240}^{PES}$  is the height of the band located at  $1240\text{ cm}^{-1}$  and attributed to PES in the pristine membrane.

**Table 2.** Description of the 3 FTIR-ATR spectrometers used in this study

**Table 3.**  $H_i/H_{1240}^{PES}$  ratio from FTIR-ATR before and after fouling of pristine membrane by proteins.

**Table 4.**  $H_i/H_{1240}^{PES}$  ratio from FTIR-ATR before and after fouling of the 210 min micro-waves aged membrane by proteins.

**Table 5.**  $H_i/H_{1240}^{PES}$  ratio from FTIR-ATR before and after fouling of the 480 min micro-waves aged membrane by proteins

**Table 6.** Determination of  $H_{1661}^{\text{protein amide I}}/H_{1240}^{PES}$  from  $(H_{1539}^{\text{protein amide II}}/H_{1240}^{PES})$  for different membranes

**Figure captions**

**Figure 1:** Some polymers constituting active layer of membranes

**Figure 2:** Opening of the PVP ring opening by hypochlorite in alkaline conditions according to Wienk et al. [22]

**Figure 3:** Evolution of PES skeleton by formation of – (a) sulfonic acid group according to [14] and (b) phenol group (OH on the phenyl ring) according to [18]

**Figure 4:** FTIR-ATR raw spectra of a pristine PES/PVP membrane (HFK-131, Koch) in green and with smaller intensity of an industrial membrane at the end of its service life (membrane U)

**Figure 5:** Raw spectra of a PES/PVP membrane - (a) without or (b) with protein fouling

**Figure 6:** Summary of the analytical difficulty to overcome and of the proposed approach

**Figure 7:** NanoSIMS analysis of the pristine HFK-131 membrane highlighting presence of nitrogen. Y-axis gives the atomic ratio of organic nitrogen to sulfur considering that N comes from PVP and S from PES. The longer is the atom ablation time on the x-axis the deeper is the analysis in the membrane. There is more PVP on the surface of the membrane but it never disappears

**Figure 8.** Scheme of the spiral membrane with 4 double sheets highlighting the different channels in which are inserted retentate or permeate spacer, respectively. Each membrane is labelled according to the same nomenclature as those used for the quantification of protein amount for the autopsied membrane (see results). F1C1 is the innermost membrane sheet whereas F4C2 is the outermost one.

**Figure 9:** Evolution of flux in skim milk according to NaOCl dose received by the spiral membrane

**Figure 10.** Mapping of the protein irreversible deposit in the spiral membrane determined from FTIR-ATR quantification (protein amount in  $\mu\text{g.cm}^{-2}$ ) according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet.

**Figure 11** Mapping of  $H_{1661}^{\text{raw spectrum}}/H_{1240}^{\text{PES}}$  obtained from FTIR-ATR according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet.

**Figure 12.** Mapping of  $H_{1661}^{\text{PVP}}/H_{1240}^{\text{PES}}$  obtained from **equation 3** according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet

**Figure 13.** Mapping of  $H_{1030}/H_{1240}^{\text{PES}}$  according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet

## APPENDICES

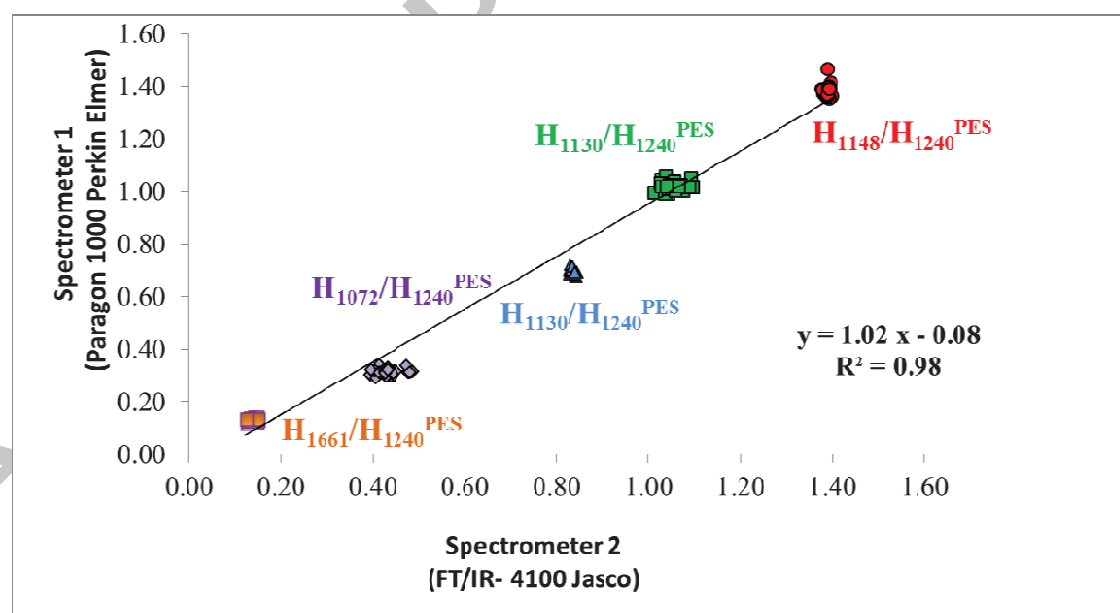
## Appendix 1: FTIR-ATR registration

One must understand that the penetration depth of the IR incident beam in the membrane depends on the thickness of the fouling deposit. Consequently direct comparison of  $H_i$  on two spectra with different fouling amounts is not possible.

To overcome this difficulty, we pay a special attention to one of the PES membrane bands, those located at  $1240\text{ cm}^{-1}$  (the height of which is further called  $H_{1240}^{\text{PES}}$ ). This band has no overlapping with PVP and protein bands. Consequently it can be used as a reference band.

We then need to determine the  $H_i/H_{1240}^{\text{PES}}$  ratio to compare variation in the quantity of the target component owing a band located at  $w_i$ .

On a practical point of view  $H_i$  has to be measured against a baseline of reference that must be chosen in a wavenumber region that is very flat and with a very small absorbance, for instance between  $2240$  and  $2060\text{ cm}^{-1}$  further called  $H_{2060-2240}^{\text{baseline}}$  with spectrometer 1.

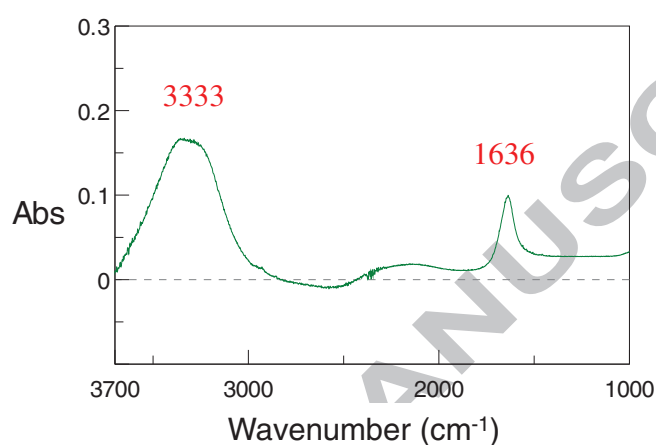


**Figure A1-1** : Linear relationship between band height ratio ( $H_i/H_{1240}^{\text{PES}}$ ) measured on different pristine HFK-131 membranes with spectrometer 1 and spectrometer 2. Similar results are obtained when comparing spectrometer 1 and spectrometer 3.

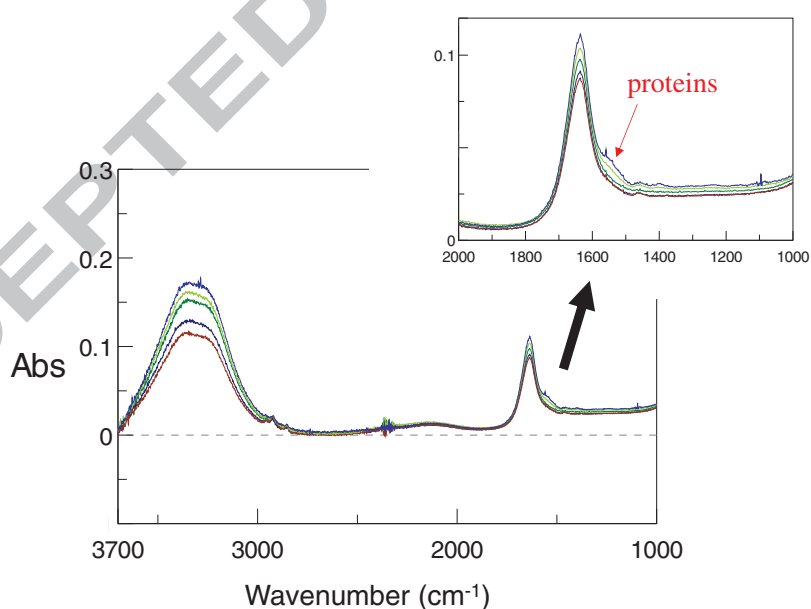
## Appendix 2: Determination of $\gamma$ for FTIR-ATR analysis of proteins in aqueous solutions.

The spectrum of water is firstly registered (**Figure A2-1 a**): two main bands due to OH bond are observed located close to  $3333\text{ cm}^{-1}$  and  $1636\text{ cm}^{-1}$ , respectively.

(a)



(b)



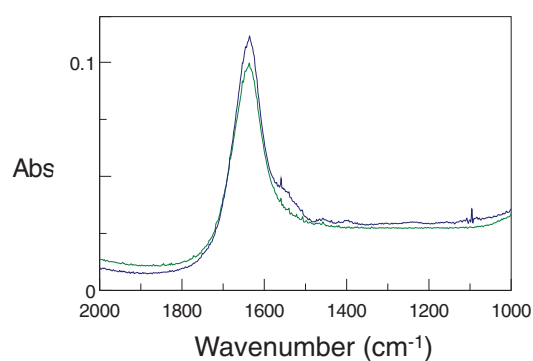
**Figure A2-1:** FTIR-ATR raw spectra- (a) water – (b) protein solutions in water

The spectra of protein solutions (protein concentration between  $5\text{ g.L}^{-1}$  and  $50\text{ g.L}^{-1}$ ) are registered (**Figure A2-1 b**). At first sight bands of water are easily observed. Looking carefully close to

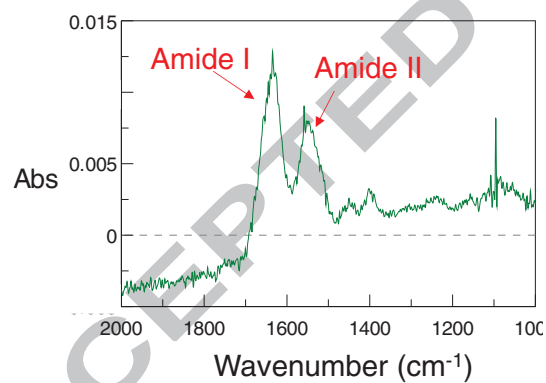
1500-1600  $\text{cm}^{-1}$ , it is shown that shoulders appear in the 1636  $\text{cm}^{-1}$  band of water. These shoulder correspond to amide I and amide II band of proteins, as it can be evidenced after substraction according to **equation A2-1** (**Figure A2-2b**).

$$\text{Protein difference spectrum} = \text{Protein raw spectrum} - \text{water spectrum} \quad (\text{eq. A2-1})$$

(a)



(b)



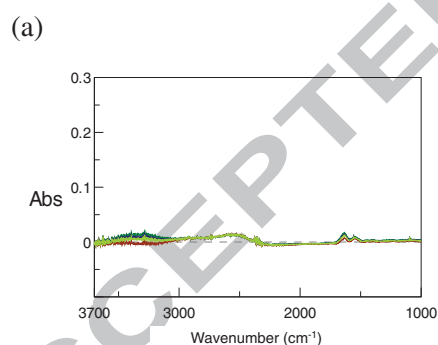


**Figure A1-2:** FTIR-ATR spectra in the 2000-1000  $\text{cm}^{-1}$  region. (a) raw spectra of water and protein solution at 50  $\text{g.L}^{-1}$ – (b) difference spectrum obtained according to **equation A2-1**.

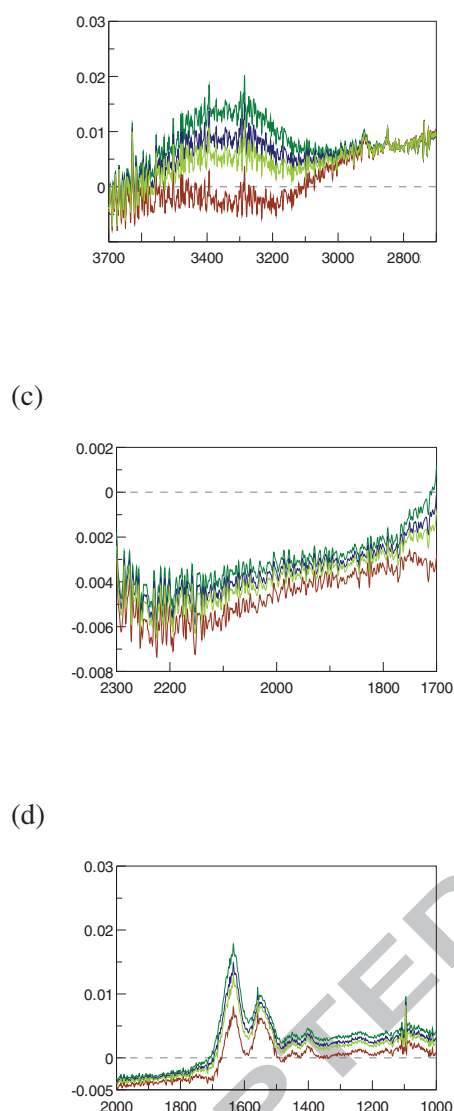
The use of **equation A2-1** is correct as a first estimation and for qualitative purpose. But in order to have a better accuracy on the height of protein bands for quantitative purpose, the equation must be modified by introducing a corrective term, further called  $\gamma$  leading at least to **equation A2-2** (similar to **eq. 6** in the full paper).

$$\text{Protein difference spectrum} = \text{Protein raw spectrum} - \gamma \text{water spectrum} \quad (\text{eq. A2-2})$$

The value of  $\gamma$  is determined from the simultaneous cancellation of both the 3333 and 1636  $\text{cm}^{-1}$  bands of the raw spectra attributed to water. **Figure A2-3** illustrates the impact of  $\gamma$  choice that is adjusted up to obtain a difference spectrum as flat as possible in the regions of interest.



(b)



**Figure A2-3:** FTIR-ATR difference spectra obtained from protein solution spectrum ( $50 \text{ g.L}^{-1}$ ) and water spectrum according to **equation A2** with various  $\gamma$  value in the range 0.95 – 1.05- (a) 3700-1000  $\text{cm}^{-1}$  range- (b) zoom in the 3700-2800  $\text{cm}^{-1}$  region to evidence cancellation of the 3333  $\text{cm}^{-1}$  water band –(c) zoom in the 2300-1700  $\text{cm}^{-1}$  region and (d) zoom in the 2000-1000  $\text{cm}^{-1}$  region to evidence cancellation of the 1661  $\text{cm}^{-1}$  water band and the appearance of amide I and amide II protein bands.

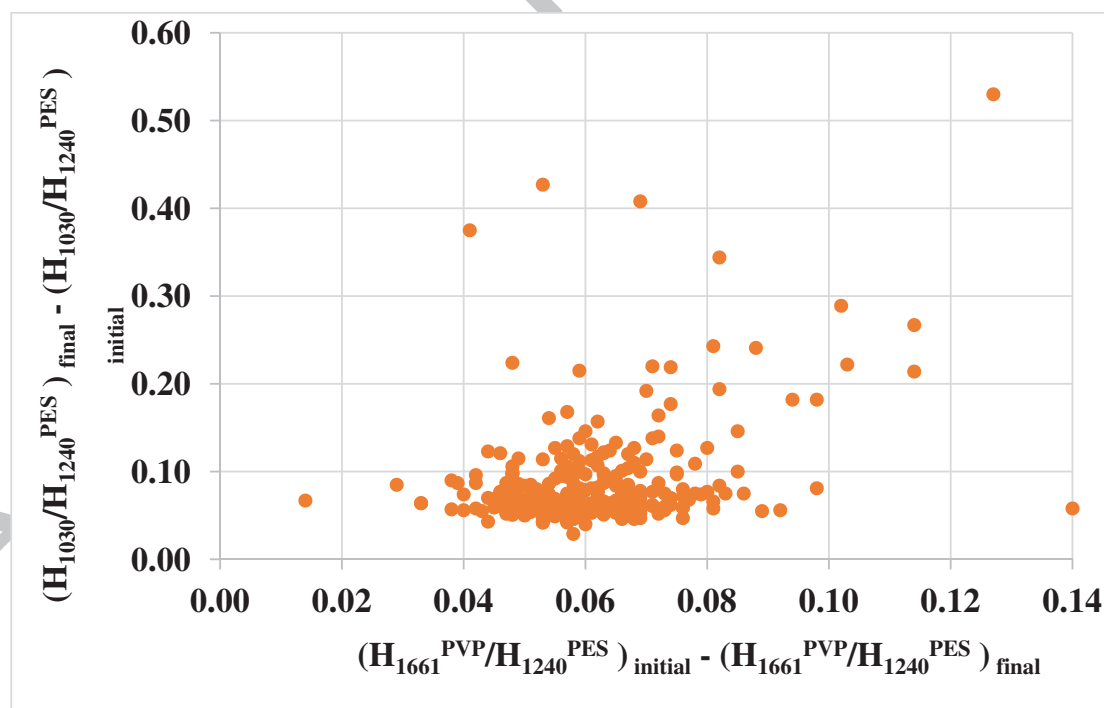
Such determination have been done for the different protein concentration. **Table A2-1** shows the optimized  $\gamma$  values for spectrum used in this study.

**Table A2-1.** Optimised  $\gamma$  value for difference spectrum (according to **equation A2**) for each protein solution. :

Protein concentration (g.L <sup>-1</sup> )	$\gamma$
5	0.920
10	0.800
20	0.699
30	0.780
50	1.018

### Appendix 3: Attempt of correlation between PVP and PES degradation

The aim is to evidence possible correlation between the disappearance of PVP and the degradation of PES (**Figure A3-1**). None is found.



**Figure A3-1:** Increase of  $1030\text{cm}^{-1}$  band (according to FTIR-ATR, pristine membrane value is 0.04) versus decrease of PVP (according to FTIR-ATR, pristine membrane value is 0.14) - (spectra acquired on 334 samples of CIP-3 spiral membrane)

**Appendix 4: FTIR-ATR supporting data corresponding to Figures 10 (A4-1), 11 (A4-2), 12 (A4-3), 13 (A4-4).**

	F1C1					F1C2			
	Inlet		Outlet			Inlet		Outlet	
	TMP(bar)					TMP(bar)			
	5.0	4.5	3.9	3.4		5.0	4.5	3.9	3.4
d (mm)	permeate axis					permeate axis			
45	68	71	57	55		59	54	57	64
135	60	58	54	49		57	59	53	51
225	58	56	58	53		54	58	53	52
315	54	61	53	49		49	43	50	52
405	52	51	53	49		48	46	50	47
495	53	52	51	48		50	57	48	44
585	51	50	49	53		62	46	42	47
675	49	55	49	46		42	52	48	61
765	47	49	48	46		47	43	42	33
855	51	48	58	47		57	34	42	38
945	37	37	42	39		25	25	26	26

	F2C1					F2C2			
	Inlet		Outlet			Inlet		Outlet	
	TMP(bar)					TMP(bar)			
	5.0	4.5	3.9	3.4		5.0	4.5	3.9	3.4
d (mm)	permeate axis					permeate axis			
45	75	64	65	53		74	73	52	55
135	49	74	53	43		68	54	55	49
225	69	56	40	46		61	62	55	56
315	55	55	39	46		53	60		47
405	46	54	47	43		54	53	50	47
495	56	48	43	47		53	49	48	46
585	44	53	50	44		50	52	48	50
675	54	50	48	46		51	50	46	42
765	45	42	41	42		47	48	42	40
855									
945									

	F3C1					F3C2			
	Inlet		Outlet			Inlet		Outlet	
	TMP(bar)					TMP(bar)			
	5.0	4.5	3.9	3.4		5.0	4.5	3.9	3.4
d (mm)	permeate axis					permeate axis			
45	82	60	58	56		63	58	51	54
135	58	56	60	52		68	53	64	50
225	69	56	52	48		41	63	58	55
315	56	53	46	44		61	56	59	53
405	52	62	46	47		54	53	49	35
495	50	52	47	48		49	70	61	50
585	49	44	47	44		54	57	51	48
675	45	49	49	48		56	51	55	36
765	45	41	44	42		49	49	50	38
855	39	41	46	44		58	41	40	42
945						13	40	40	29

	F4C1					F4C2			
	Inlet		Outlet			Inlet		Outlet	
	TMP(bar)					TMP(bar)			
	5.0	4.5	3.9	3.4		5.0	4.5	3.9	3.4
d (mm)	permeate axis					permeate axis			
45	71	54	61	54		73	66	52	50
135	62	62	66	49		59	65	51	48
225	57	60	64	48		63	41	46	49
315	53	48	61	47		48	47	44	50
405	53	51	53	45		43	54	70	46
495	47	51	53	48		46	49	45	35
585	49	51	50	51		40	45	46	42
675	47	52	48	45		49	43	49	37
765	49	40	48	46		48	36	73	31
855	46	45	42	37		47	44	44	32
945									

**Figure A4-1.** Mapping of the protein irreversible deposit in the spiral membrane determined from FTIR-ATR quantification (protein amount in  $\mu\text{g}\cdot\text{cm}^{-2}$ ) according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. (white = not determined)

F1C1					F1C2				
d (mm)	Inlet	Outlet			Inlet	Outlet			
	TMP(bar)				TMP(bar)				
	5.0	4.5	3.9	3.4	5.0	4.5	3.9	3.4	
	permeate axis					permeate axis			
45	0.306	0.316	0.268	0.257	0.283	0.272	0.242	0.261	
135	0.269	0.274	0.252	0.248	0.283	0.280	0.263	0.258	
225	0.276	0.267	0.274	0.251	0.268	0.279	0.268	0.244	
315	0.248	0.259	0.255	0.235	0.235	0.222	0.257	0.230	
405	0.251	0.249	0.260	0.239	0.237	0.221	0.254	0.244	
495	0.261	0.256	0.255	0.232	0.236	0.255	0.245	0.226	
585	0.242	0.250	0.242	0.248	0.234	0.242	0.206	0.241	
675	0.240	0.250	0.239	0.235	0.194	0.254	0.238	0.221	
765	0.236	0.243	0.242	0.233	0.215	0.219	0.237	0.173	
855	0.253	0.237	0.258	0.234	0.117	0.205	0.218	0.196	
945	0.197	0.202	0.219	0.194	0.139	0.150	0.158	0.150	

F2C1					F2C2				
Inlet					Inlet				
TMP(bar)					TMP(bar)				
5.0					5.0				
4.5					4.5				
3.9					3.9				
3.4					3.4				
d (mm)	permeate axis				d (mm)	permeate axis			
45	0.333	0.297	0.289	0.253	45	0.303	0.289	0.248	0.250
135	0.250	0.336	0.260	0.228	135	0.295	0.266	0.263	0.252
225	0.264	0.267	0.215	0.234	225	0.287	0.281	0.252	0.248
315	0.257	0.251	0.204	0.214	315	0.255	0.286	?	0.239
405	0.233	0.259	0.240	0.221	405	0.260	0.254	0.248	0.240
495	0.256	0.232	0.222	0.228	495	0.264	0.252	0.245	0.239
585	0.221	0.256	0.237	0.211	585	0.255	0.195	0.247	0.240
675	0.255	0.249	0.239	0.233	675	0.259	0.255	0.243	0.223
765	0.220	0.214	0.217	0.214	765	0.242	0.244	0.222	0.206
855					855				
945					945				

F3C1					F3C2				
Inlet					Inlet				
TMP(bar)					TMP(bar)				
5.0					5.0				
4.5					4.5				
3.9					3.9				
3.4					3.4				
d (mm)	permeate axis				d (mm)	permeate axis			
45	0.335	0.302	0.261	0.270	45	0.272	0.270	0.240	0.267
135	0.275	0.271	0.279	0.248	135	0.309	0.274	0.272	0.255
225	0.285	0.275	0.268	0.248	225	0.262	0.281	0.280	0.261
315	0.262	0.259	0.233	0.243	315	0.271	0.267	0.262	0.246
405	0.257	0.268	0.235	0.231	405	0.259	0.269	0.247	0.230
495	0.246	0.263	0.242	0.240	495	0.259	0.288	0.265	0.253
585	0.243	0.231	0.244	0.233	585	0.256	0.257	0.259	0.248
675	0.233	0.235	0.245		675	0.251	0.250	0.261	0.223
765	0.256	0.221	0.228	0.222	765	0.242	0.247	0.254	0.227
855	0.218	0.220	0.234	0.229	855	0.271	0.227	0.235	0.220
945					945	0.138	0.205	0.217	0.172

F4C1					F4C2				
Inlet					Inlet				
TMP(bar)					TMP(bar)				
5.0					5.0				
4.5					4.5				
3.9					3.9				
3.4					3.4				
d (mm)	permeate axis				d (mm)	permeate axis			
45	0.316	0.265	0.283	0.239	45	0.274	0.290	0.255	0.251
135	0.275	0.278	0.264	0.246	135	0.275	0.251	0.204	0.235
225	0.266	0.273	0.297	0.239	225	0.289	0.213	0.235	0.251
315	0.255	0.235	0.274	0.228	315	0.248	0.237	0.212	0.231
405	0.257	0.257	0.240	0.231	405	0.213	0.266	0.272	0.221
495	0.244	0.253	0.265	0.234	495	0.230	0.217	0.216	0.189
585	0.243	0.240	0.244	0.228	585	0.211	0.225	0.227	?
675	0.232	0.249	0.220	0.215	675	0.229	0.225	0.232	0.186
765	0.256	0.206	0.231	0.243	765	0.250	0.205	0.244	0.192
855	0.227	0.225	0.199	0.191	855	0.220	0.201	0.226	0.176
945					945				

**Figure A4-2** Mapping of  $H_{1661}^{\text{raw spectra}}/H_{1240}^{\text{PES}}$  obtained from FTIR-ATR according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. (white = not determined)

	F1C1					F1C2			
	Inlet		Outlet			Inlet		Outlet	
	TMP(bar)					TMP(bar)			
	5.0	4.5	3.9	3.4		5.0	4.5	3.9	3.4
d (mm)	permeate axis					permeate axis			
45	0.090	0.091	0.085	0.079		0.092	0.098	0.059	0.058
135	0.078	0.089	0.077	0.086		0.099	0.089	0.091	0.092
225	0.089	0.085	0.086	0.078		0.093	0.092	0.094	0.073
315	0.073	0.063	0.082	0.075		0.074	0.079	0.095	0.062
405	0.069	0.048	0.075	0.078		0.081	0.069	0.092	0.090
495	0.088	0.086	0.088	0.075		0.072	0.071	0.089	0.080
585	0.078	0.087	0.082	0.076		0.037	0.085	0.067	0.087
675	0.080	0.073	0.079	0.082		0.055	0.086	0.081	0.026
765	0.081	0.083	0.084	0.081		0.060	0.077	0.096	0.060
855	0.088	0.079	0.072	0.080		0	0.088	0.077	0.069
945	0.071	0.077	0.080	0.064		0.051	0.059	0.064	0.058

	F2C1					F2C2			
	Inlet		Outlet			Inlet		Outlet	
	TMP(bar)					TMP(bar)			
	5.0	4.5	3.9	3.4		5.0	4.5	3.9	3.4
d (mm)	permeate axis					permeate axis			
45	0.096	0.093	0.081	0.082		0.068	0.058	0.079	0.073
135	0.091	0.102	0.087	0.087		0.078	0.089	0.086	0.092
225	0.046	0.086	0.081	0.082		0.092	0.083	0.074	0.067
315	0.079	0.073	0.072	0.061		0.083	0.092		0.085
405	0.083	0.083	0.087	0.079		0.085	0.081	0.085	0.084
495	0.076	0.074	0.081	0.072		0.091	0.091	0.087	0.086
585	0.075	0.084	0.074	0.066		0.091	0.026	0.089	0.077
675	0.081	0.086	0.083	0.082		0.092	0.091	0.090	0.084
765	0.072	0.072	0.079	0.074		0.086	0.087	0.081	0.073
855									
945									

	F3C1					F3C2			
	Inlet		Outlet			Inlet		Outlet	
	TMP(bar)					TMP(bar)			
	5.0	4.5	3.9	3.4		5.0	4.5	3.9	3.4
d (mm)	permeate axis					permeate axis			
45	0.077	0.107	0.075	0.090		0.070	0.082	0.073	0.092
135	0.089	0.088	0.086	0.078		0.094	0.101	0.068	0.093
225	0.066	0.095	0.097	0.090		0.126	0.081	0.094	0.083
315	0.082	0.088	0.080	0.096		0.076	0.086	0.072	0.075
405	0.087	0.071	0.084	0.078		0.085	0.098	0.087	0.111
495	0.082	0.093	0.086	0.083		0.098	0.065	0.071	0.089
585	0.083	0.086	0.090	0.087		0.082	0.072	0.093	0.092
675	0.085	0.075	0.084			0.071	0.083	0.082	0.100
765	0.107	0.085	0.083	0.081		0.081	0.086	0.091	0.100
855	0.087	0.083	0.082	0.083		0.085	0.089	0.102	0.079
945						0.082	0.071	0.082	0.069

F4C1					F4C2						
	Inlet		Outlet				Inlet		Outlet		
	TMP(bar)		4.5	3.9	3.4		TMP(bar)		4.5	3.9	3.4
	5.0						5.0				
d (mm)	permeate axis						permeate axis				
45	0.092	0.088	0.089		0.064		0.042	0.080	0.086		0.086
135	0.081	0.055	0.087		0.081		0.086	0.042	0.038		0.078
225	0.082	0.092	0.083		0.083		0.087	0.075	0.082		0.090
315	0.077	0.078	0.073		0.084		0.090	0.081	0.065		0.068
405	0.090	0.069	0.083		0.090		0.071	0.092	0.052		0.070
495	0.087	0.092	0.075		0.084		0.080	0.057	0.066		0.069
585	0.074	0.080	0.062		0.078		0.078	0.077	0.075		
675	0.080	0.064	0.066		0.094		0.068	0.081	0.071		0.060
765	0.071	0.074	0.091		0.077		0.092	0.083	0.013		0.085
855	0.077	0.059	0.065		0.077		0.067	0.054	0.080		0.064
945											



**Figure A4-3.** Mapping of  $H_{1661}^{PVP}/H_{1240}^{PES}$  obtained from **equation 3** according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. (white = not determined)

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F1 C1					F1 C2				
Inlet		Outlet			Inlet		Outlet		
TMP (bar)					TMP (bar)				
5.0		4.5	3.9	3.4	5.0		4.5	3.9	3.4
d (mm)	permeate axis				d (mm)	permeate axis			
45	0.124	0.126	0.108	0.121	45	0.264	0.136	0.283	0.234
135	0.110	0.112	0.106	0.092	135	0.415	0.125	0.155	0.117
225	0.111	0.105	0.114	0.118	225	0.127	0.138	0.113	0.144
315	0.103	0.108	0.109	0.096	315	0.141	0.171	0.099	0.149
405	0.101	0.096	0.100	0.100	405	0.121	0.178	0.098	0.092
495	0.101	0.103	0.099	0.107	495	0.167	0.448	0.094	0.119
585	0.096	0.098	0.116	0.097	585	0.262	0.132	0.115	0.088
675	0.103	0.097	0.100	0.088	675	0.140	0.104	0.255	0.307
765	0.099	0.108	0.101	0.109	765	0.167	0.162	0.110	0.117
855	0.104	0.104	0.102	0.104	855	0.098	0.120	0.093	0.117
945	0.087	0.091	0.092	0.112	945	0.095	0.098	0.087	0.124

F2 C1					F2 C2				
Inlet		Outlet			Inlet		Outlet		
TMP (bar)					TMP (bar)				
5.0		4.5	3.9	3.4	5.0		4.5	3.9	3.4
d (mm)	permeate axis				d (mm)	permeate axis			
45	0.163	0.119	0.178	0.127	45	0.204	0.384	0.153	0.124
135	0.124	0.130	0.104	0.084	135	0.122	0.120	0.114	0.096
225	0.222	0.094	0.152	0.096	225	0.116	0.114	0.121	0.100
315	0.100	0.096	0.098	0.114	315	0.144	0.111		0.091
405	0.091	0.134	0.089	0.098	405	0.100	0.119	0.098	0.105
495	0.104	0.115	0.091	0.086	495	0.100	0.094	0.091	0.110
585	0.124	0.135	0.091	0.110	585	0.100	0.254	0.101	0.161
675	0.099	0.091	0.105	0.086	675	0.095	0.098	0.095	0.098
765	0.098	0.111	0.093	0.086	765	0.108	0.099	0.097	0.092
855					855				
945					945				

F3 C1					F3 C2				
Inlet		Outlet			Inlet		Outlet		
TMP (bar)					TMP (bar)				
5.0		4.5	3.9	3.4	5.0		4.5	3.9	3.4
d (mm)	permeate axis				d (mm)	permeate axis			
45	0.127	0.104	0.135	0.106	45	0.232	0.136	0.125	0.146
135	0.096	0.106	0.099	0.117	135	0.117	0.127	0.127	0.105
225	0.217	0.109	0.095	0.098	225	0.107	0.141	0.101	0.115
315	0.111	0.099	0.080	0.083	315	0.164	0.201	0.150	0.098
405	0.114	0.103	0.101	0.102	405	0.109	0.127	0.116	0.125
495	0.092	0.092	0.090	0.091	495	0.098	0.137	0.112	0.098
585	0.094	0.095	0.090	0.087	585	0.098	0.148	0.094	0.091
675	0.090	0.098	0.093		675	0.110	0.208	0.119	0.096
765	0.104	0.089	0.085	0.094	765	0.107	0.109	0.094	0.114
855	0.082	0.082	0.090	0.084	855	0.101	0.105	0.097	0.097
945					945	0.069	0.118	0.086	0.102

F4 C1					F4 C2				
Inlet		Outlet			Inlet		Outlet		
TMP (bar)					TMP (bar)				
5.0		4.5	3.9	3.4	5.0		4.5	3.9	3.4
d (mm)	permeate axis				d (mm)	permeate axis			
45	0.131	0.116	0.122	0.120	45	0.222	0.137	0.110	0.111
135	0.140	0.186	0.467	0.097	135	0.126	0.121	0.329	0.157
225	0.160	0.137	0.169	0.101	225	0.154	0.173	0.103	0.099
315	0.128	0.147	0.160	0.141	315	0.102	0.101	0.164	0.180
405	0.106	0.101	0.146	0.096	405	0.096	0.139	0.281	0.154
495	0.082	0.110	0.093	0.155	495	0.186	0.115	0.102	0.260
585	0.102	0.119	0.115	0.197	585	0.116	0.136	0.130	
675	0.103	0.099	0.259	0.161	675	0.092	0.119	0.140	0.116
765	0.093	0.096	0.111	0.092	765	0.091	0.083	0.570	0.167
855	0.098	0.106	0.139	0.138	855	0.096	0.115	0.096	0.099
945					945				

**Figure A4-4.** Mapping of  $H_{1030}/H_{1240}^{PES}$  according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. (white = not determined).

## Tables

**Table 1.** Conditions of membrane accelerated ageing, FTIR-ATR characteristics and protein amount further adsorbed on (aged) membrane surface.

	Ageing time (min) (micro-waves)	FTIR-ATR* $H_{1661}^{PVP}/H_{1240}^{PES}$	Range of protein deposit on membrane surface ( $\mu\text{g}\cdot\text{cm}^{-2}$ )
Pristine membrane	0	$0.14 \pm 0.01$ (n=6)	0 – 208
Ageing by NaOCl	210	$0.10 \pm 0.01$ (n=6)	0 - 76
Ageing by NaOCl	480	$0.07 \pm 0.01$ (n=6)	0- 82

\*  $H_{1661}^{PVP}$  is the height of the band located at  $1661\text{ cm}^{-1}$  and attributed to PVP in the pristine membrane whereas  $H_{1240}^{PES}$  is the height of the band located at  $1240\text{ cm}^{-1}$  and attributed to PES in the pristine membrane.

**Table 2.** Description of the three FTIR-ATR spectrometers used in this study

	<b>Spectrometer 1</b>	<b>Spectrometer 2</b>	<b>Spectrometer 3</b>
	Paragon 1000	FT/IR-4100	Spectrum 100
	Perkin-Elmer	Jasco	Perkin Elmer
ATR	ZnSe crystal	ZnSe crystal	Diamond crystal
accessory	incidence angle: 45°	incidence angle: 45°	incidence angle: 45°
	12 reflections	1 reflection	1 reflection
Crystal dimension	2 cm x 5 cm	1.8 mm diameter	2 mm diameter
Software	Spectrum for windows 5	Spectra Manager II	Spectrum for windows 6
Acquisition conditions	spectral domain : 4000-600 cm <sup>-1</sup>	spectral domain : 3700-600 cm <sup>-1</sup>	spectral domain : 4000-600 cm <sup>-1</sup>
	resolution : 2 cm <sup>-1</sup>	resolution : 2 cm <sup>-1</sup>	resolution : 2 cm <sup>-1</sup>
	20 scans	128 scans	20 scans
	background : air	background : air	background : air

**Table 3.**  $H_i/H_{1240}^{PES}$  ratio from FTIR-ATR before and after fouling of pristine membrane by proteins.

Protein concentration in solution (g.L <sup>-1</sup> )	$H_{1661}^{PVP}/H_{1240}^{PES}$ (pristine)	$H_{1661}^{raw\ spectra}/H_{1240}^{PES}$ (fouled)	$H_{1539}^{protein\ amide\ II}/H_{1240}^{PES}$ (fouled)	$H_{1661}^{protein\ amide\ I}/H_{1240}^{PES}$ (from Eq. 3)
0	0.152			
5	0.137	0.202	0.084	0.065
10	0.125	0.264	0.150	0.139
20	0.148	0.321	0.197	0.174
30	0.146	0.633	0.568	0.487
50	0.134	0.742	0.723	0.608

**Table 4**–  $H_i/H_{1240}^{PES}$  ratio from FTIR-ATR before and after fouling of the 210 min micro-waves aged membrane by proteins.

Protein concentration in solution (g.L <sup>-1</sup> )	$H_{1661}^{PVP}/H_{1240}^{PES}$ (not fouled)	$H_{1661}^{raw\ spectra}/H_{1240}^{PES}$ (fouled)	$H_{1539}^{protein\ amide}/H_{1240}^{PES}$ (fouled)	$H_{1661}^{protein\ amide}/H_{1240}^{PES}$ (from Eq. 3)
0	0.116			
5	0.084	0.149	0.084	0.065
10	0.093	0.167	0.094	0.075
20	0.093	0.198	0.126	0.105
30	0.100	0.290	0.213	0.189
50	0.099	0.348	0.275	0.248

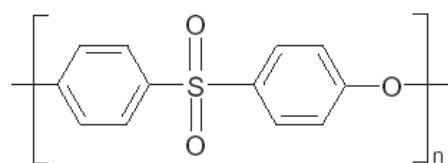
**Table 5** –  $H_i/H_{1240}^{PES}$  ratio from FTIR-ATR before and after fouling of the 480 min micro-waves aged membrane by proteins

Protein concentration in solution (g.L <sup>-1</sup> )	$H_{1661}^{PVP}/H_{1240}^{PES}$ (not fouled)	$H_{1661}^{raw\ spectra}/H_{1240}^{PES}$ (fouled)	$H_{1539}^{protein\ amide}/H_{1240}^{PES}$ (fouled)	$H_{1661}^{protein\ amide}/H_{1240}^{PES}$ (from Eq. 3)
0	0.073			
5	0.058	0.113	0.060	0.055
10	0.060	0.154	0.098	0.094
20	0.066	0.208	0.144	0.141
30	0.076	0.252	0.177	0.176
50	0.062	0.361	0.297	0.299

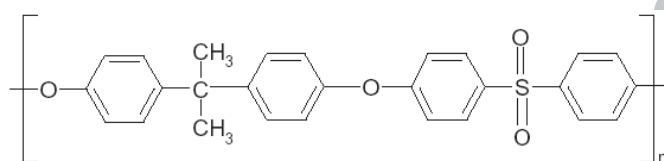
**Table 6-** Determination of  $H_{1661}^{\text{protein amide I}}/H_{1240}^{\text{PES}}$  from  $(H_{1539}^{\text{protein amide II}}/H_{1240}^{\text{PES}})$  for different membranes

$(H_{1661}^{\text{PVP}}/H_{1240}^{\text{PES}})$	$H_{1661}^{\text{protein amide I}}/H_{1240}^{\text{PES}} = a$ $(H_{1539}^{\text{protein amide II}}/H_{1240}^{\text{PES}}) (y = a x)$	Number of values	Correlation coefficient $R^2$	Range of protein amount on membrane surface ( $\mu\text{g.cm}^{-2}$ )
0.14 (Pristine)	$y = 0.85 x$	6	0.999	0 - 208
0.10 (210 min aged)	$y = 0.88 x$	6	0.975	0 - 76
0.07 (480 min aged)	$y = 0.99 x$	6	0.996	0 - 82
<b>Average</b>	<b><math>y = 0.87 x</math></b>	<b>18</b>	<b>0.991</b>	<b>0 - 76 (208)</b>

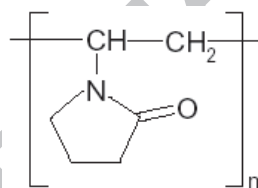
## Figures



(a) Polyethersulfone



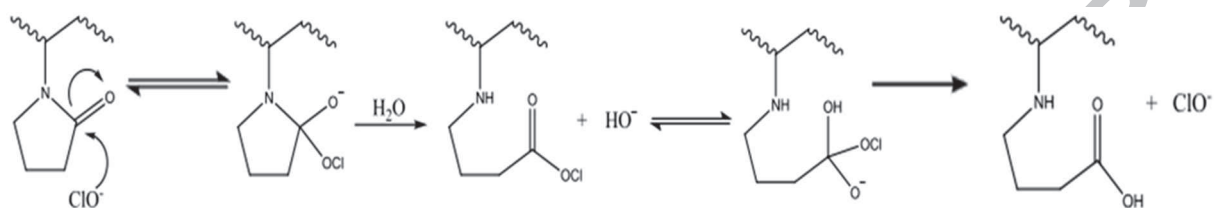
(b) Polysulfone



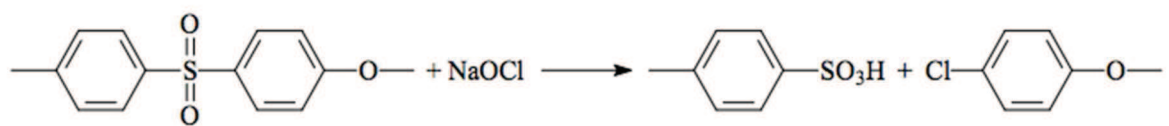
(c) Polyvinylpyrrolidone

**Figure 1:** Some polymers constituting active layer of membranes

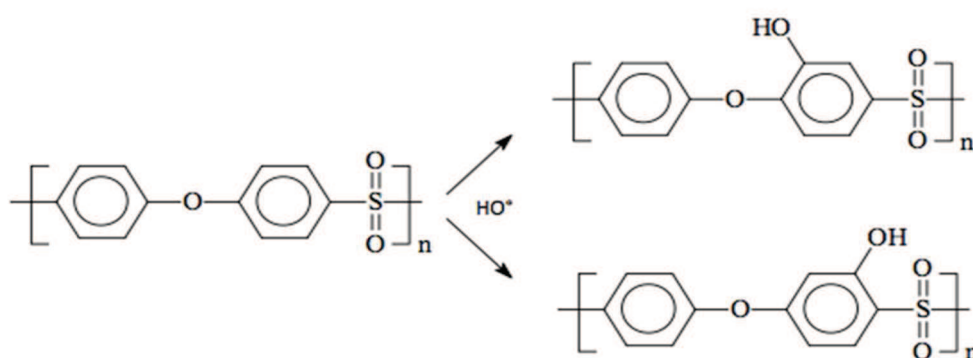




**Figure 2:** Opening of the PVP ring opening by hypochlorite in alkaline conditions according to Wienk et al. [22]

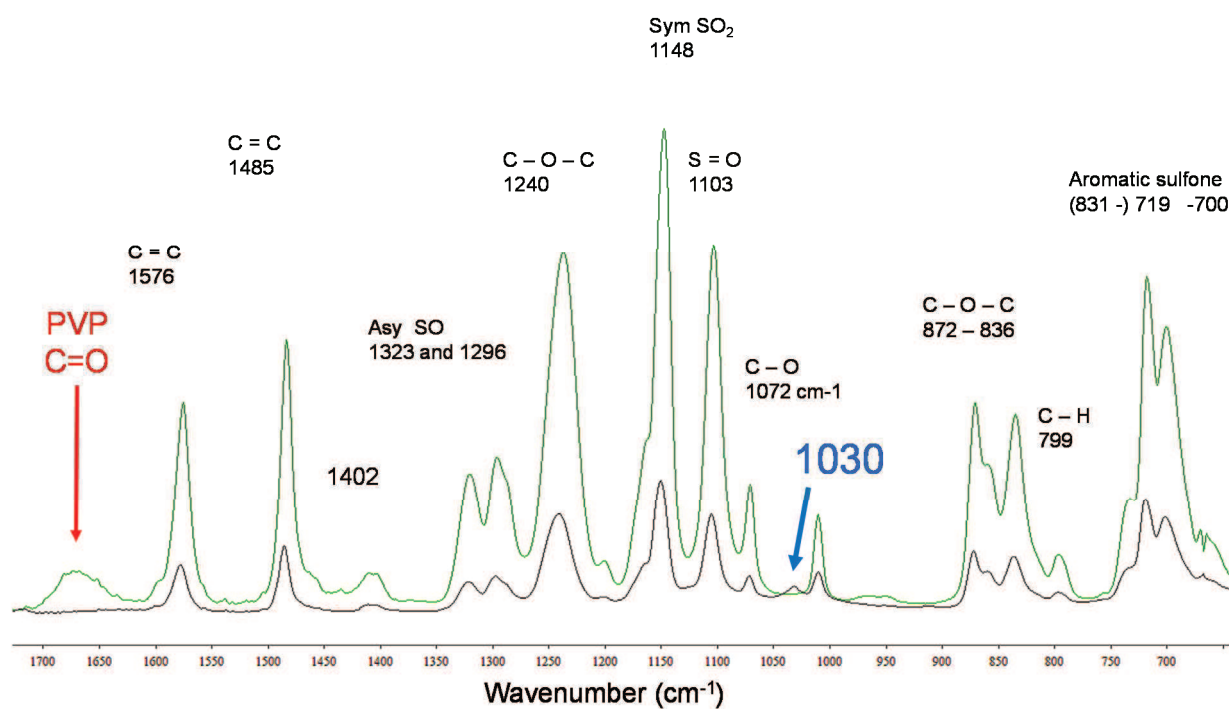


(a)

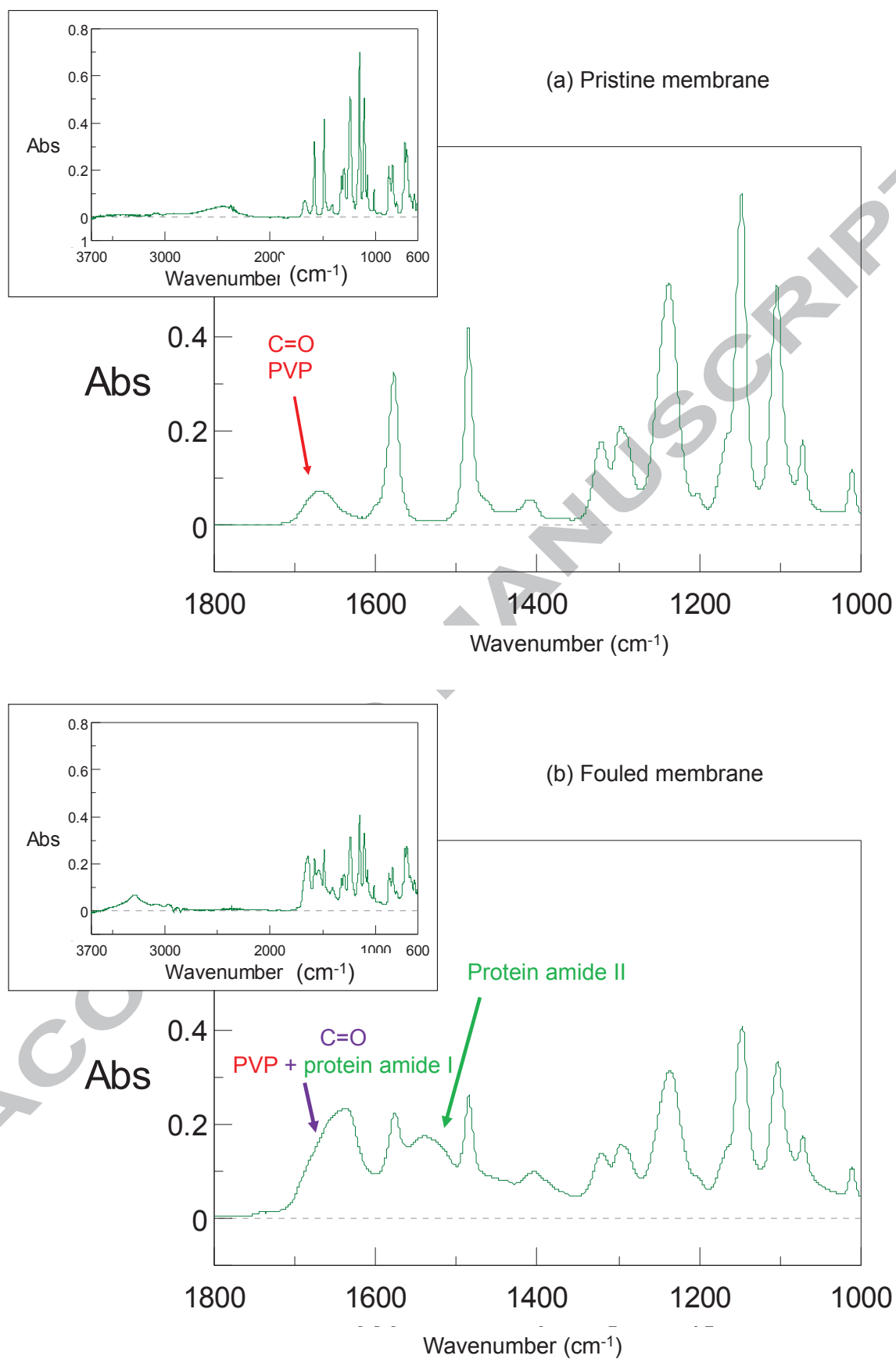


(b)

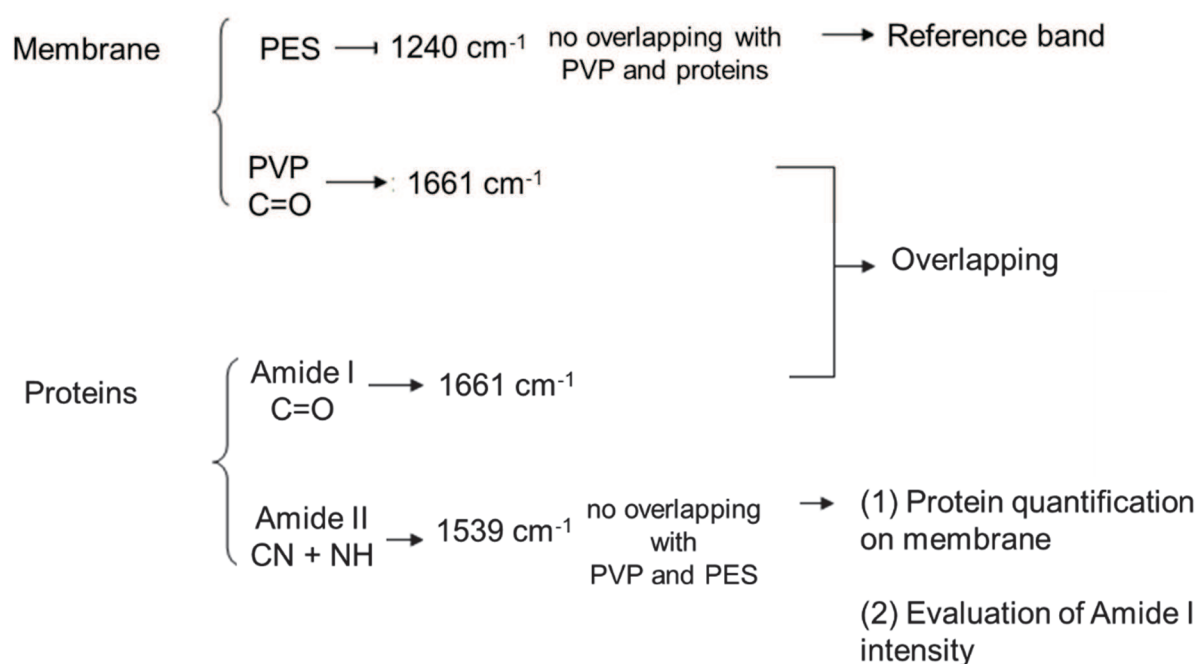
**Figure 3:** Evolution of PES skeleton by formation of – (a) sulfonic acid group according to [14] and (b) phenol group (OH on the phenyl ring) according to [18]



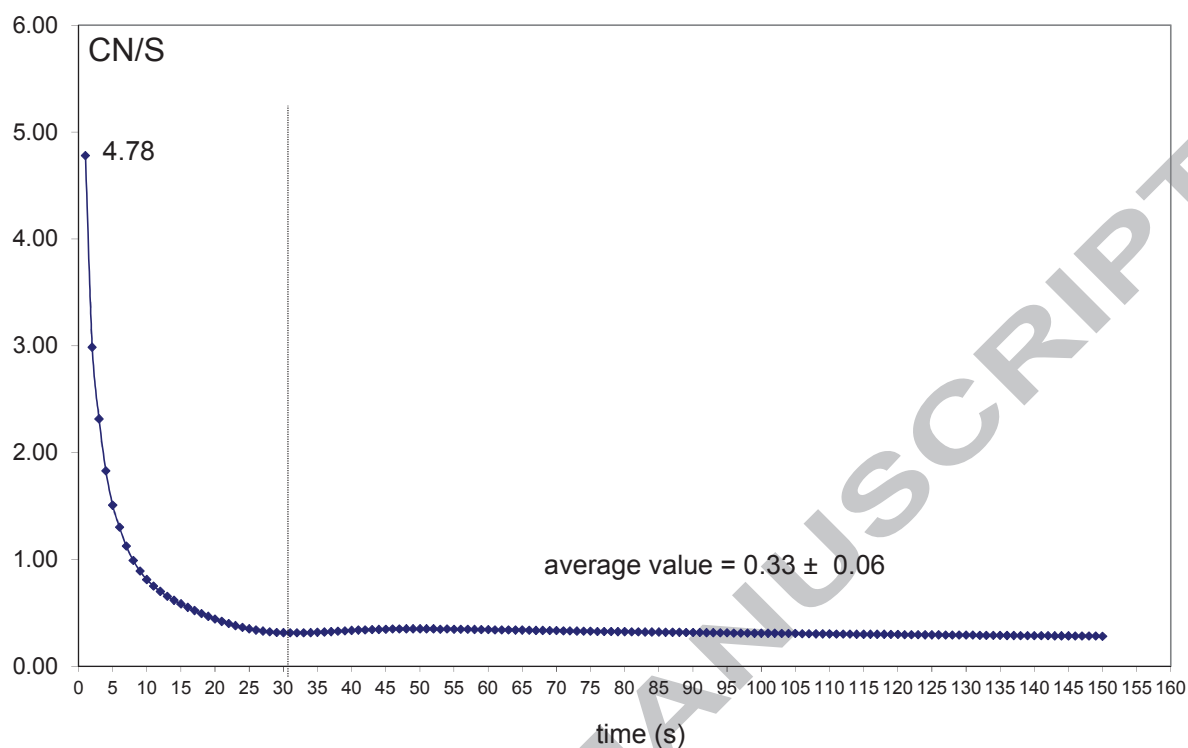
**Figure 4:** FTIR-ATR raw spectra of a pristine PES/PVP membrane (HFK-131, Koch) in green and with smaller intensity of an industrial membrane at the end of its service life (membrane U)



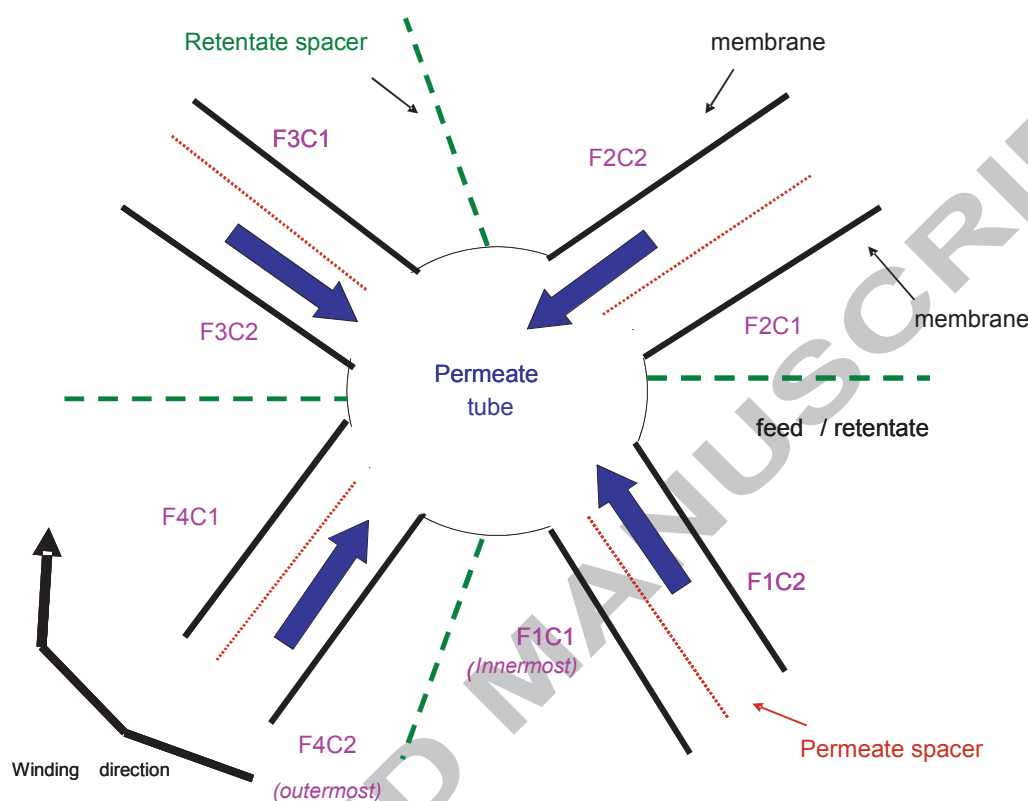
**Figure 5:** Raw spectra of a PES/PVP membrane - (a) without or (b) with protein fouling



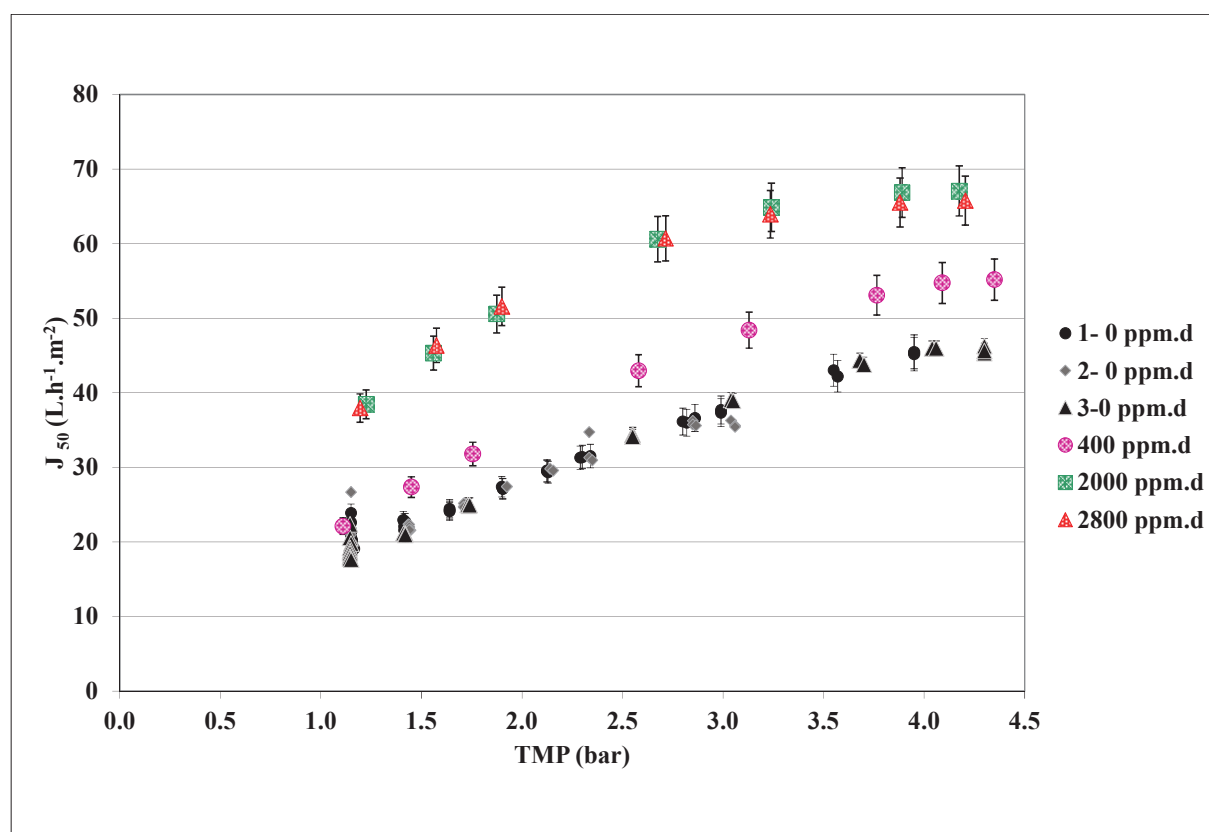
**Figure 6:** Summary of the analytical difficulty to overcome and of the proposed approach



**Figure 7:** NanoSIMS analysis of the pristine HFK-131 membrane highlighting presence of nitrogen. Y-axis gives the atomic ratio of organic nitrogen to sulfur considering that N comes from PVP and S from PES. The longer is the atom ablation time on the x-axis the deeper is the analysis in the membrane. There is more PVP on the surface of the membrane but it never disappears

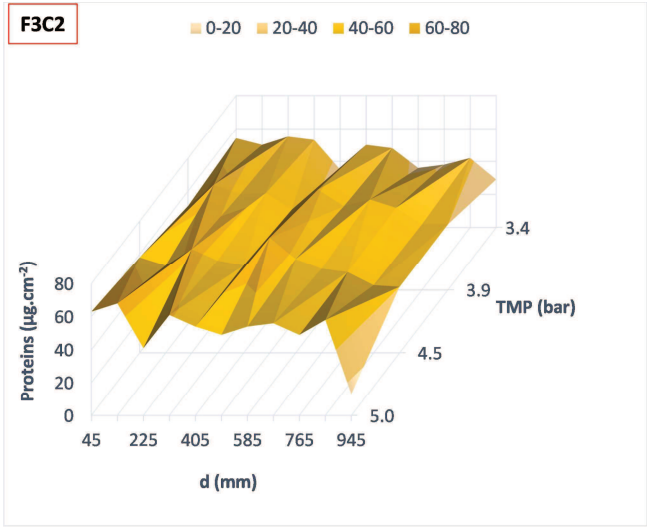
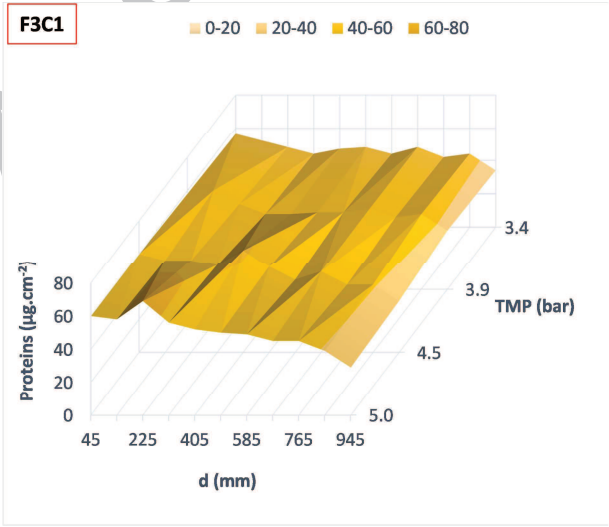
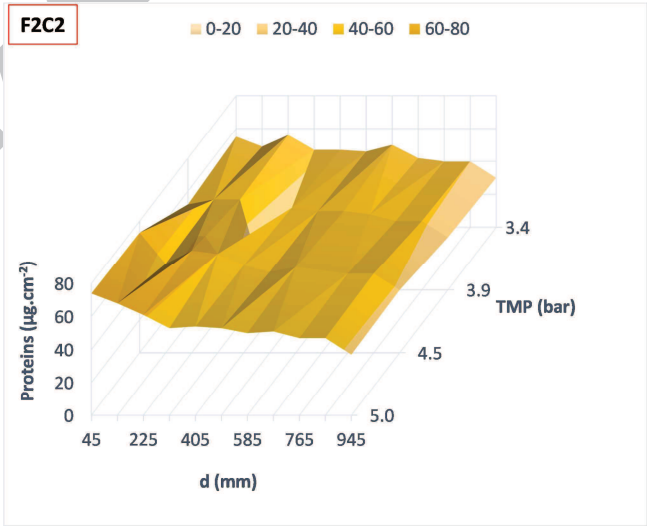
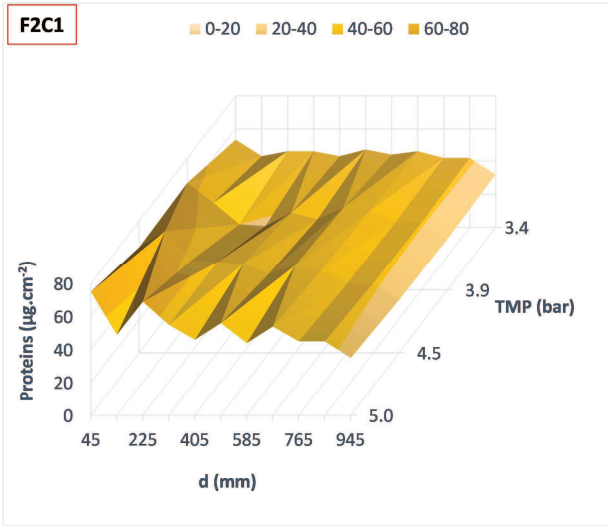
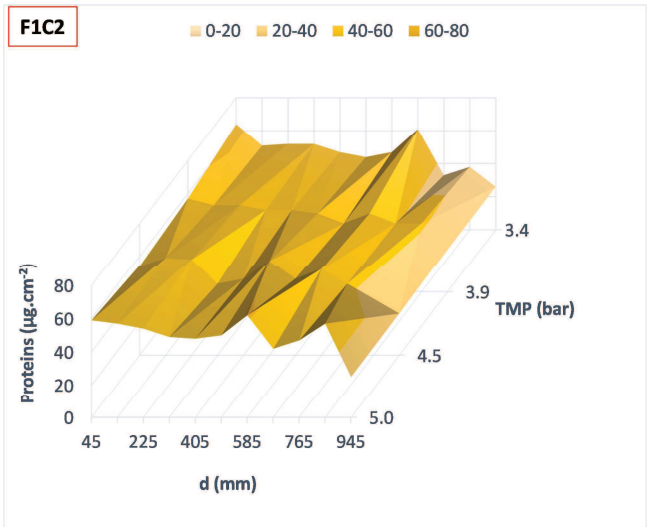
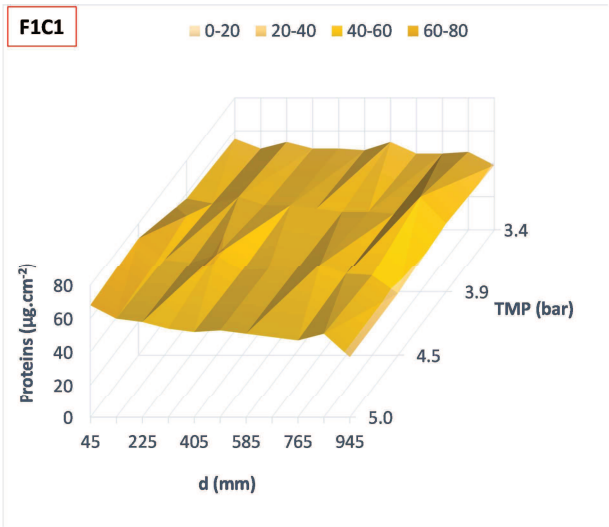


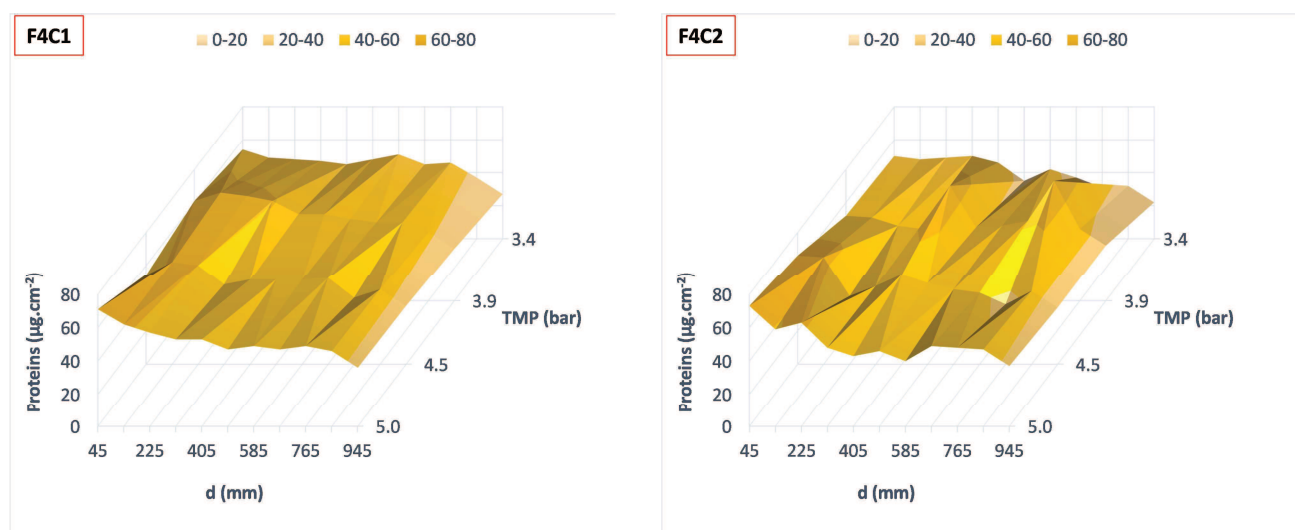
**Figure 8.** Scheme of the spiral membrane with 4 double sheets highlighting the different channels in which are inserted retentate or permeate spacer, respectively. Each membrane is labelled according to the same nomenclature as those used for the quantification of protein amount for the autopsied membrane (see results). F1C1 is the innermost membrane sheet whereas F4C2 is the outermost one.



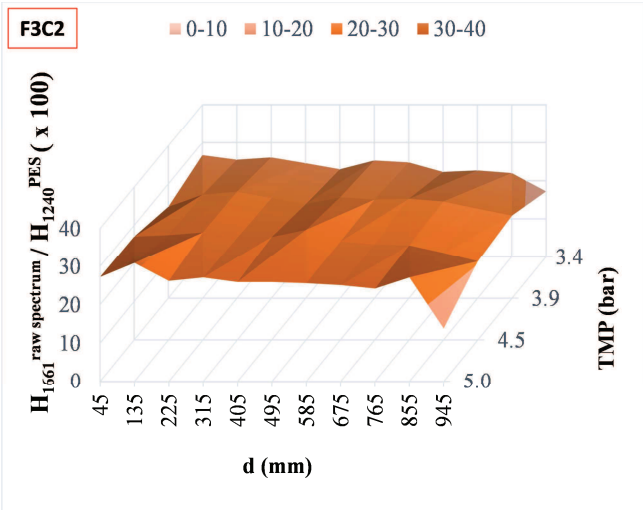
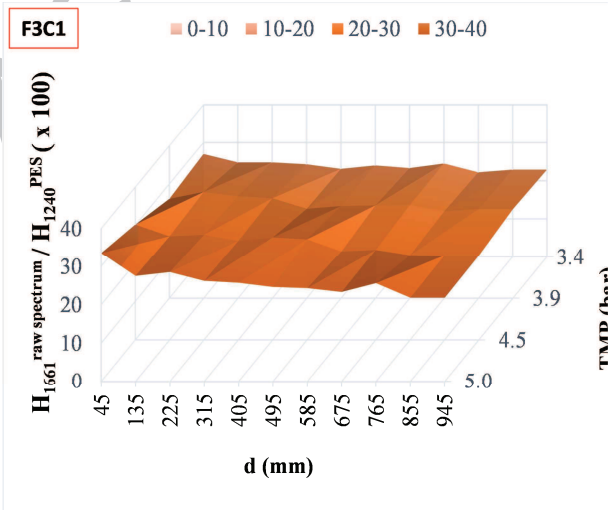
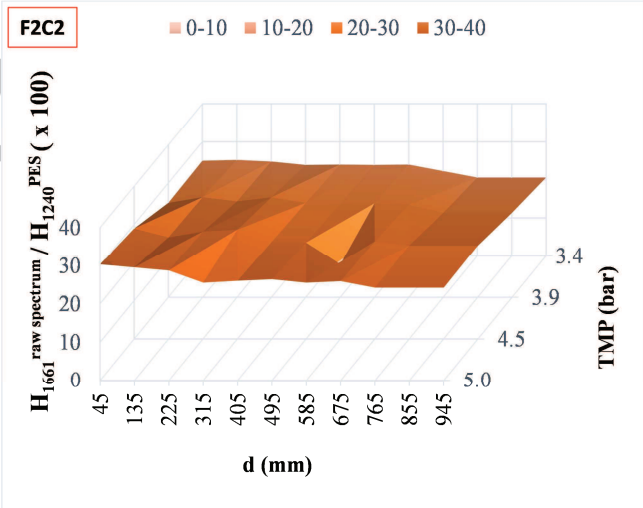
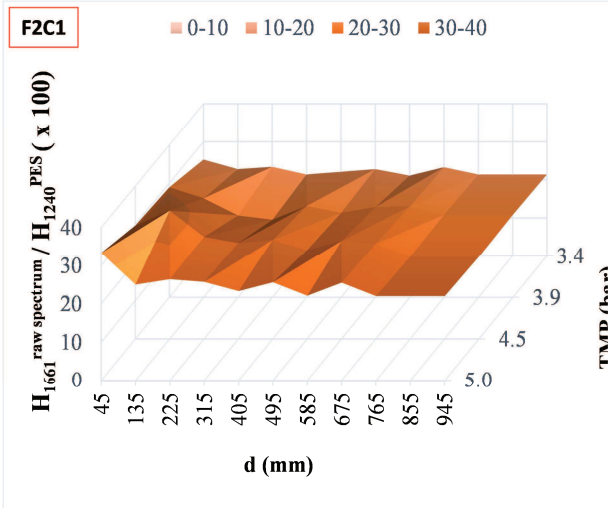
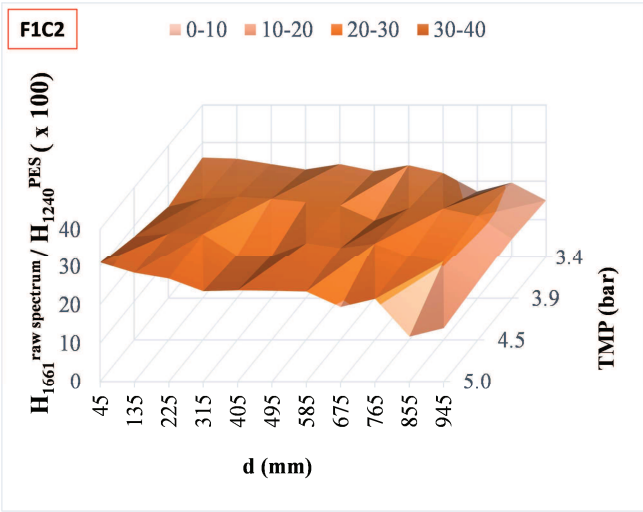
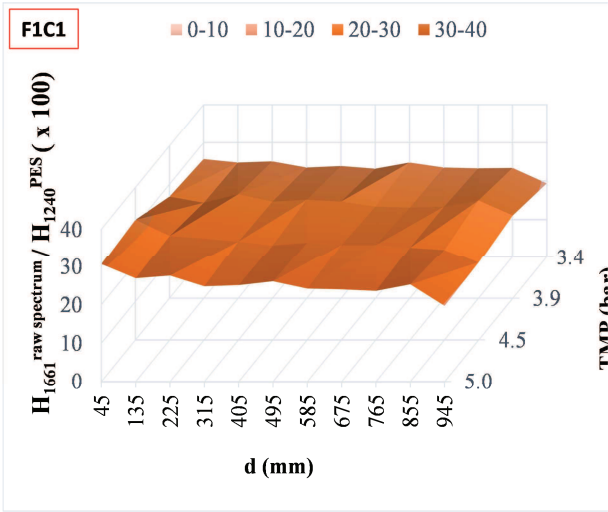
**Figure 9:** Evolution of flux in skim milk according to NaOCl dose received by the spiral membrane

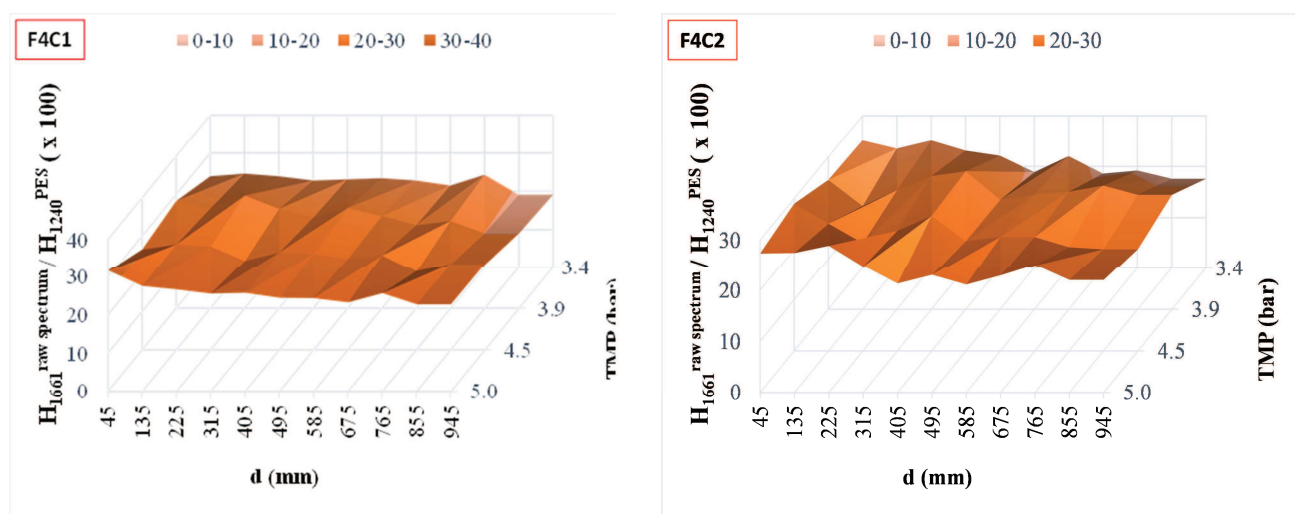




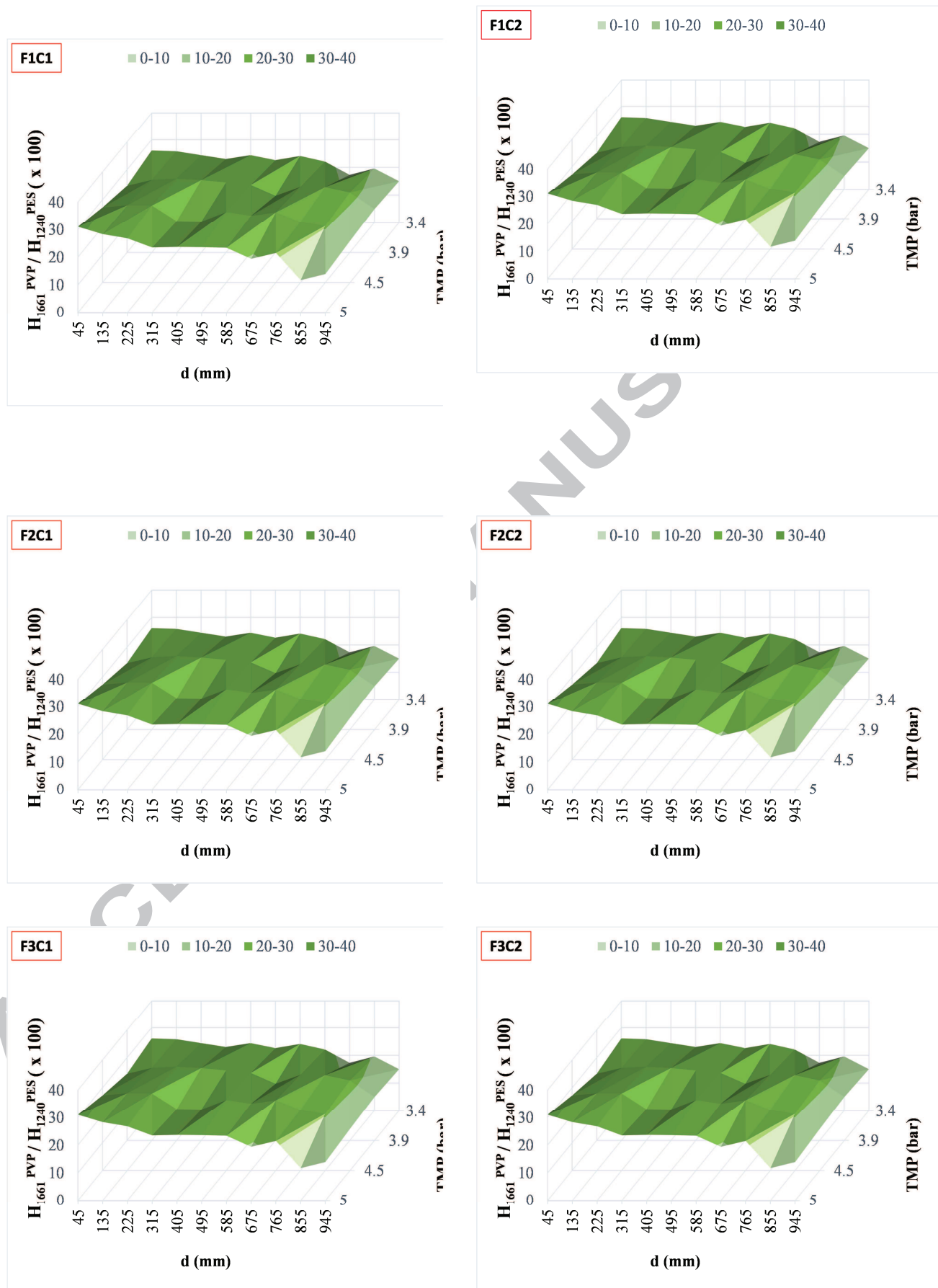


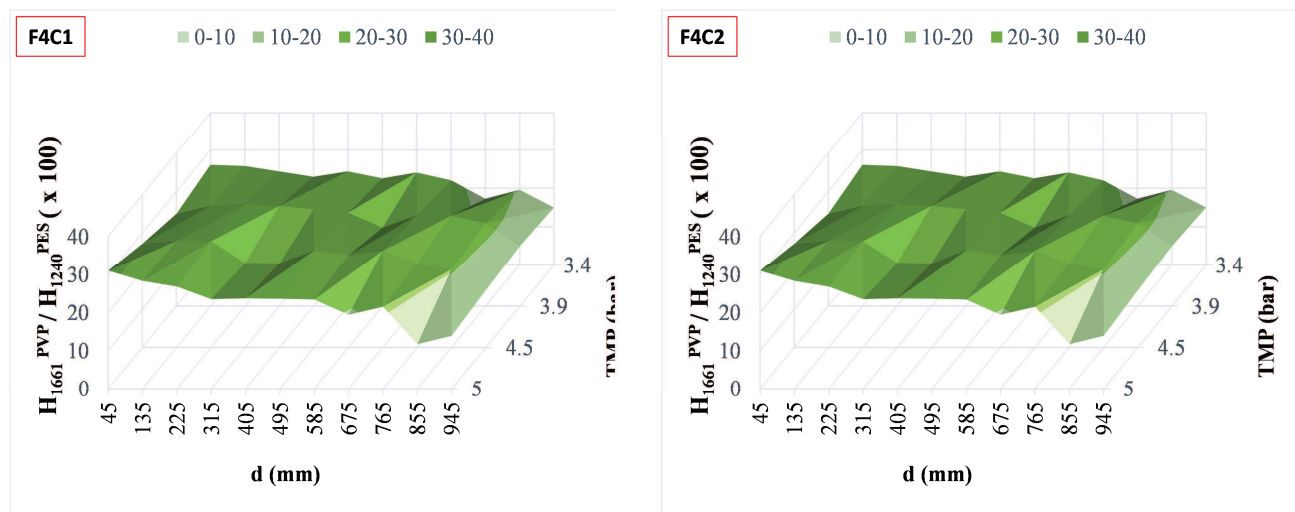
**Figure 10.** Mapping of the protein irreversible deposit in the spiral membrane determined from FTIR-ATR quantification (protein amount in  $\mu\text{g.cm}^{-2}$ ) according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet.





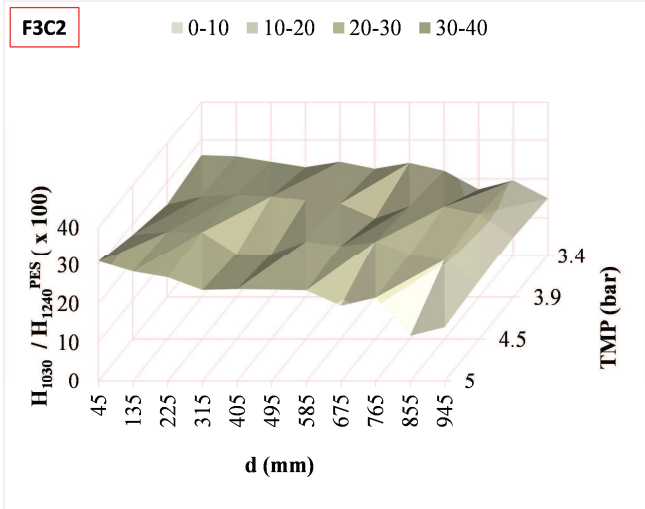
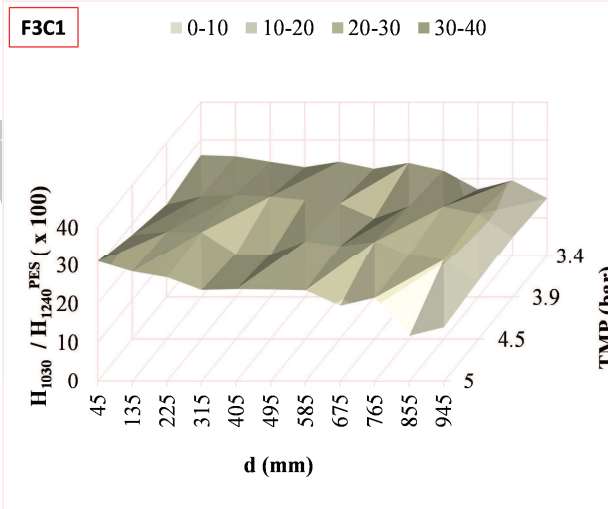
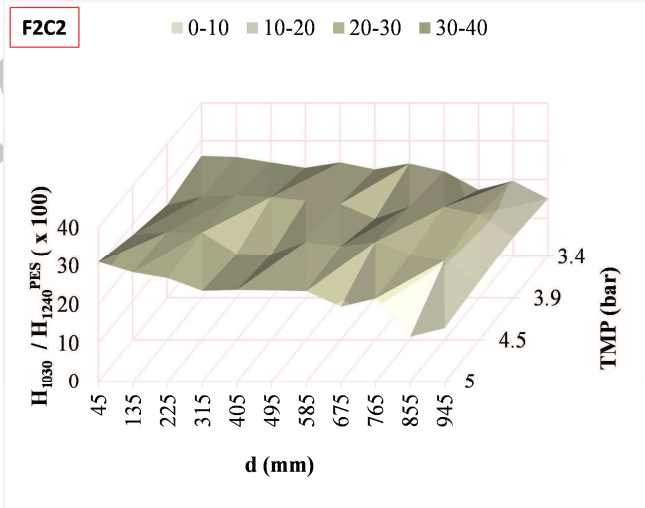
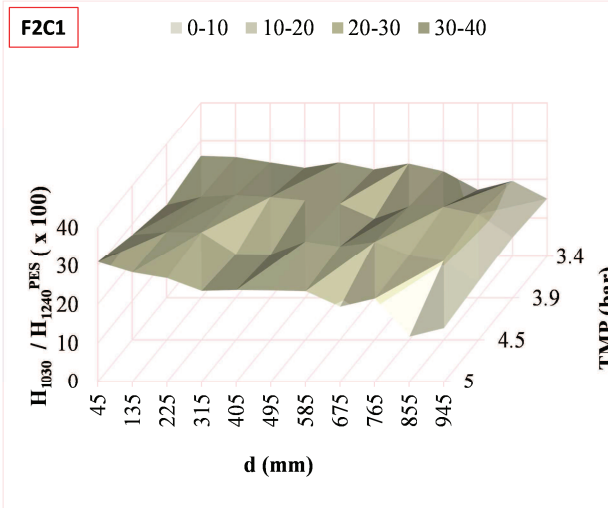
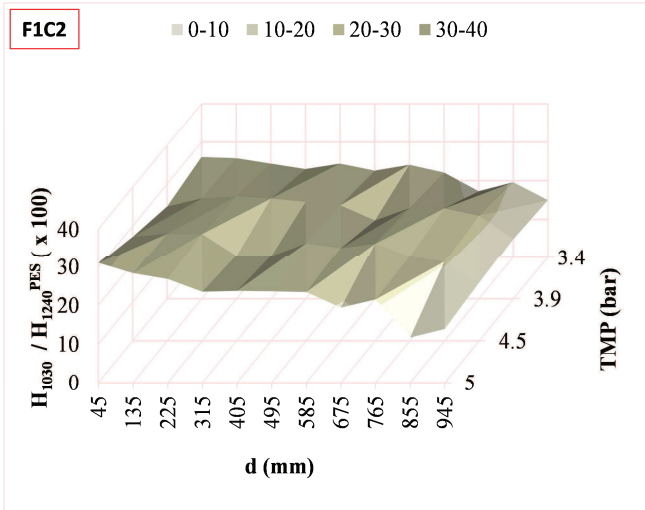
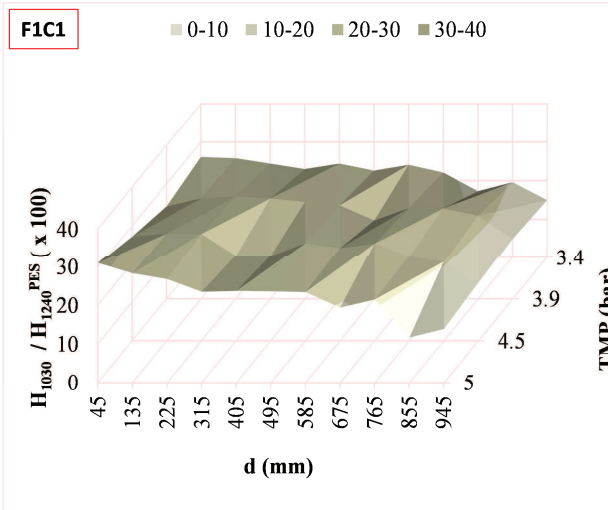
**Figure 11** Mapping of  $H_{1661}^{\text{raw spectrum}} / H_{1240}^{\text{PES}}$  obtained from FTIR-ATR according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet.

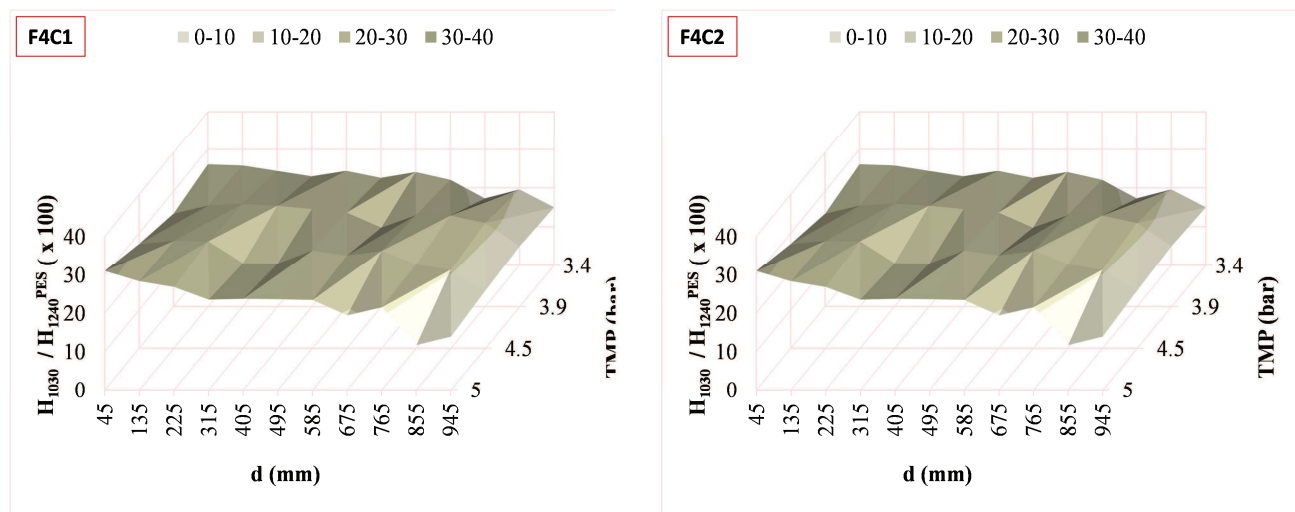




**Figure 12.** Mapping of  $H_{1661}^{PVP} / H_{1240}^{PES}$  obtained from **equation 3** according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet







**Figure 13.** Mapping of  $H_{1030}/H_{1240}^{PES}$  according to the location in the spiral membrane. The local TMP is calculated from the assumption of a linear pressure drop decrease. The membrane labels are defined on **Figure 8**. TMP decreases from the inlet to the outlet of the spiral membrane.  $d$  is the distance from the permeate axis ( $d=0$ ) for a membrane sheet



## Highlights

- - evidencing PES/PVP membrane degradation by FTIR despite presence of protein fouling
- 
- - methodology of treatment of FTIR spectra to reveal membrane degradation due to NaOCl
- 
- - application to mapping of PVP degradation due to NaOCl in a spiral membrane
- 
- - application to mapping of PES degradation due to NaOCl in a spiral membrane